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UNITED STATES DISTRICT COURT DISTRICT OF NEW JERSEY

INDIVIOR INC., INDIVIOR UK LIMITED, and AQUESTIVE THERAPEUTICS, INC.,

Plaintiffs,

V .

PAR PHARMACEUTICAL, INC., PAR PHARMACEUTICAL COMPANIES INC., and ENDO INTERNATIONAL PLC,

Defendants.

Civil Action No.			
(Filed Electronically)			

COMPLAINT FOR PATENT INFRINGEMENT

Plaintiffs Indivior Inc. (formerly known as Reckitt Benckiser Pharmaceuticals Inc.)

("Indivior"), Indivior UK Limited (formerly known as RB Pharmaceuticals Limited) ("Indivior UK"), and Aquestive Therapeutics, Inc. (formerly known as MonoSol Rx LLC) ("Aquestive")

(collectively, "Plaintiffs") file this Complaint against Defendants Par Pharmaceutical, Inc. ("Par Pharmaceutical"), Par Pharmaceutical Companies Inc. ("Par Pharmaceutical Cos.") and Endo International PLC ("Endo") (collectively, "Par" or "Defendants") and allege as follows:

NATURE OF THE ACTION

1. This is an action for patent infringement arising under the Food and Drug Laws and Patent Laws of the United States, Titles 21 and 35 of the United States Code, respectively, arising from Par's submission of an Abbreviated New Drug Application ("ANDA") to the Food and Drug Administration ("FDA") seeking approval to manufacture, use, and sell a generic version of Plaintiffs' Suboxone[®] sublingual film prior to the expiration of United States Patent No. 9,855,221 ("the '221 patent" or "the patent-in-suit").

THE PARTIES

- 2. Plaintiff Indivior is a Delaware corporation having a principal place of business at 10710 Midlothian Turnpike, Suite 430, Richmond, Virginia.
- 3. Plaintiff Indivior UK is a United Kingdom corporation having a principal place of business at 103-105 Bath Road, Slough, UK.
- 4. Plaintiff Aquestive Therapeutics, Inc. is a Delaware limited liability corporation having a principal place of business at 30 Technology Drive, Warren, New Jersey 07059.
- 5. On information and belief, Defendant Par Pharmaceutical is a corporation organized and existing under the laws of New York, having a principal place of business at One Ram Ridge Road, Spring Valley, New York 10977.
- 6. On information and belief, Par Pharmaceutical is a wholly owned subsidiary of Par Pharmaceutical Cos.
- 7. On information and belief, Par Pharmaceutical is a wholly owned subsidiary, directly or indirectly, of Endo and holds itself out as "an Endo International Company."
- 8. On information and belief, Defendant Par Pharmaceutical Cos. is a corporation organized and existing under the laws of Delaware, having a principal place of business at One Ram Ridge Road, Spring Valley, New York 10977.

- 9. On information and belief, Defendant Par Pharmaceutical Cos. is a holding company.
- 10. On information and belief, Defendant Par Pharmaceutical Cos. is a wholly owned subsidiary, directly or indirectly, of Endo.
- 11. On information and belief, Defendant Endo International PLC is a publicly-traded company organized and existing under the laws of Ireland, having a place of business at First Floor, Minerva House, Simmonscourt Road, Ballsbridge, Dublin 4, Ireland.
- 12. On information and belief, Defendant Endo has a regular and established place of business within this Judicial District in Cranbury, New Jersey.

JURISDICTION AND VENUE

- 13. This Court has subject matter jurisdiction over this action pursuant to 28 U.S.C. §§ 1331, 1338(a), 2201, and 2202.
- 14. This Court has personal jurisdiction over Par Pharmaceutical because of, *inter alia*, Par Pharmaceutical's continuous and systematic contacts with corporate entities within this Judicial District, its previous submission to the jurisdiction of this Judicial District, and its substantial, continuous, and systematic distribution, marketing, and/or sales of generic pharmaceutical products to residents of this Judicial District.
- 15. This Court has personal jurisdiction over Par Pharmaceutical Cos. because of, *inter alia*, its continuous and systematic contacts with the State of New Jersey and corporate entities within this Judicial District, including its subsidiary, agent, and/or alter-ego, Par Pharmaceutical, a company registered with the State of New Jersey's Department of Health as a drug manufacturer and wholesaler, its previous submission to the jurisdiction of this Judicial District, and its substantial, continuous, and systematic distribution, marketing, and/or sales of

generic pharmaceutical products to residents of this Judicial District including through, directly or indirectly, Par Pharmaceutical.

- 16. This Court has personal jurisdiction over Endo because of, *inter alia*, Endo's continuous and systematic contacts with corporate entities within this Judicial District, and Endo's marketing and sales activities in this Judicial District, including, but not limited to, the substantial, continuous, and systematic distribution, marketing, and/or sales of generic pharmaceutical products to residents of this Judicial District.
- 17. On information and belief, Par Pharmaceutical is in the business of, *inter alia*, developing, manufacturing, obtaining regulatory approval for, marketing, selling, and distributing pharmaceutical products, including generic copies of branded pharmaceuticals, in New Jersey and throughout the United States.
- 18. On information and belief, Par Pharmaceutical directly or indirectly manufactures, markets, and sells generic drug products throughout the United States and in this Judicial District.
- 19. Upon information and belief, Par Pharmaceutical is registered with the State of New Jersey's Department of Health as a "Manufacturer and Wholesale[r]," registration number 5004032.
- 20. On information and belief, Par Pharmaceutical has availed itself of the jurisdiction of this Court by previously filing lawsuits in this Judicial District. *See, e.g., Par Pharm., Inc. et al. v. Luitpold Pharms., Inc.*, No. 16-1190 (D.N.J.); *Par Pharm., Inc. v. Breckenridge Pharm., Inc.*, No. 13-4000 (D.N.J.).
- 21. On information and belief, Par Pharmaceutical and Par Pharmaceutical Cos. have previously been sued in this Judicial District and have not challenged personal jurisdiction and

venue. See, e.g., Horizon Therapeutics, LLC v. Par Pharmaceutical, Inc., No. 17-5901 (D.N.J.); Celgene Corp. v. Par Pharmaceutical Inc., No. 17-03159 (D.N.J.); Alcon Labs., Inc. v. Dr. Reddy's Labs., No. 16-6775 (D.N.J.); BioMarin Pharmaceutical Inc. v. Par Pharmaceutical Inc., No. 16-01015 (D.N.J.); Jazz Pharms., Inc. v. Par Pharm., Inc., No. 15-7580 (D.N.J.); Shire LLC v. Par Pharm., Inc., No. 15-1454 (D.N.J.); Supernus Pharms., Inc. v. Par Pharm. Cos., Inc., No. 15-326 (D.N.J.).

22. On information and belief, Par Pharmaceutical and Par Pharmaceutical Cos. have further availed themselves of the jurisdiction of this Court by filing counterclaims in this Judicial District. See, e.g., Horizon Therapeutics, LLC v. Par Pharmaceutical, Inc., No. 17-5901 (D.N.J.); Alcon Laboratories, Inc. v. Dr. Reddy's Laboratories Ltd. et al., No. 16-6775 (D.N.J.); West-Ward Pharmaceuticals Corp. v. Par Pharmaceuticals Inc., No. 16-5456 (D.N.J.); Horizon Therapeutics, LLC v. Par Pharmaceutical Inc., No. 16-3910 (D.N.J.); Merck Sharp & Dohme Corp. v. Par Sterile Products, LLC, No. 16-948 (D.N.J.); BioMarin Pharmaceutical Inc. v. Par Pharmaceutical Inc., No. 16-1015 (D.N.J.); Fresenius Kabi USA, LLC v. Fera Pharmaceuticals, LLC, No. 15-3654 (D.N.J.); Jazz Pharmaceuticals, Inc. v. Par Pharmaceutical Inc., No. 15-7580 (D.N.J.); BioMarin Pharmaceutical Inc. et al. v. Par Pharmaceutical Inc., No. 15-1706 (D.N.J.); Alcon Pharmaceuticals Ltd. v. Par Pharmaceutical Inc., No. 15-7240 (D.N.J.); Fresenius Kabi USA, LLC v. Par Sterile Products, LLC, No. 15-3852 (D.N.J.); Jazz Pharmaceuticals, Inc. v. Amneal Pharmaceuticals, LLC, 15-3217 (D.N.J.); Helsinn Healthcare S.A. v. Par Pharmaceutical Companies Inc., No. 15-2078 (D.N.J.); Shire LLC v. Par Pharmaceutical Inc., No. 15-1454 (D.N.J.); Supernus Pharmaceuticals, Inc. v. Par Pharmaceutical Companies, Inc., No. 15-326 (D.N.J.); Jazz Pharmaceuticals, Inc. v. Par Pharmaceutical Inc., No. 15-173 (D.N.J.); Jazz Pharmaceuticals, Inc. v. Par Pharmaceutical, Inc., No. 14-6150 (D.N.J.); Jazz

Pharmaceuticals, Inc. v. Par Pharmaceutical, Inc., No. 14-5139 (D.N.J.); Jazz Pharmaceuticals, Inc. v. Par Pharmaceutical, Inc., No. 13-7884 (D.N.J.); Purdue Pharmaceutical Products L.P. v. Par Pharmaceutical, Inc., No. 12-6738 (D.N.J.); Depomed, Inc. v. Impax Laboratories, Inc., No. 12-2154 (D.N.J.).

23. On information and belief, Endo acquired Par Pharmaceutical Holdings, Inc. on September 25, 2015. Endo's 2015 Form 10-K states, "Immediately following the closing, Par Pharmaceutical Holdings, Inc. changed its name to Par Pharmaceutical Companies, Inc. (Par)." *See* Endo 2015 Form 10-K at 3,

https://www.sec.gov/Archives/edgar/data/1593034/000159303415000005/endp-12312014x10k.htm (last visited February 7, 2018). Endo's 2015 Form 10-K also states that "Par has operated in two business segments, (i) Par Pharmaceutical, which includes generic products ... and (ii) Par Specialty Pharmaceuticals, which markets three branded products." *Id*.

- 24. On information and belief, Par Pharmaceutical holds itself out as "an Endo International Company." In the Paragraph IV Notice Letter Par Pharmaceutical sent to Plaintiffs regarding the '221 patent, Par Pharmaceutical identified itself as a "Endo International Company."
- 25. On information and belief, Endo holds out to the public that it has a physical location in Cranbury, New Jersey. *See, e.g.*, http://www.endo.com/about-us/locations (last visited February 7, 2018).
- 26. On information and belief, Endo operates and maintains a regular and established place of business located at 7 Clarke Drive, Cranbury, New Jersey 08512. *See, e.g.*, http://www.endo.com/about-us/locations (last visited February 7, 2018).

- 27. On information and belief, Endo lists a Research and Development property, located in Cranbury, New Jersey, as part of its "U.S. Generic Pharmaceuticals Segment" on its most recent Form 10-K filed on March 1, 2017. *See* Endo 2017 Form 10-K at 43, https://www.sec.gov/Archives/edgar/data/1593034/000159303417000009/endp-12312016x10k.htm (last visited February 7, 2018).
- 28. On information and belief, Par Pharmaceutical and Par Pharmaceutical Cos. work in concert with respect to the regulatory approval, manufacturing, marketing, sale, and distribution of generic pharmaceutical products throughout the United States, including in this Judicial District.
- 29. On information and belief, Par Pharmaceutical acts at the direction, and for the benefit, of Par Pharmaceutical Cos. and Endo, and is controlled and/or dominated by Par Pharmaceutical Cos. and Endo.
- 30. On information and belief, Endo, either directly or indirectly through its wholly owned subsidiaries, is in the business of making and selling generic pharmaceutical products, which it distributes, markets, and/or sells in New Jersey and throughout the United States.
- 31. On information and belief, Par Pharmaceutical, Par Pharmaceutical Cos., and Endo hold themselves out as a unitary entity for purposes of manufacturing, marketing, selling, and distributing generic pharmaceutical products in the United States.
- 32. Aquestive has legitimate and significant reasons for bringing this action in this District. For example, Aquestive's principal place of business is in this District. Aquestive's primary witnesses are in, or are regularly in, this District, as is the bulk of Aquestive's evidence and records currently in its possession, custody, or control. Defendants' course of conduct is designed to cause the performance of the tortious act of patent infringement that has led to

foreseeable harm and injury to Aquestive, which is a New Jersey corporation. Further,

Defendants would not suffer hardship, undue or otherwise, by being summoned to this District.

33. Venue is proper in this District under 28 U.S.C. §§ 1391 and 1400.

THE PATENT-IN-SUIT

34. Plaintiff Aquestive Therapeutics, Inc. (formerly known as MonoSol Rx LLC) is the lawful owner of the '221 patent, and Plaintiff Indivior is an exclusive licensee of the '221 patent and holds the exclusionary rights to market and sell Suboxone® sublingual film in the United States. The '221 patent, entitled, "Uniform Films for Rapid-Dissolve Dosage Form Incorporating Anti-Tacking Compositions," was duly and legally issued on January 2, 2018, naming Garry L. Myers, Pradeep Sanghavi, Andrew Philip Verrall, Vimala Francis, and Laura Brooks as inventors. A true copy of the '221 patent is attached hereto as Exhibit A.

SUBOXONE® SUBLINGUAL FILM

- 35. Plaintiff Indivior is the holder of New Drug Application ("NDA") No. 22-410 for Suboxone® (buprenorphine hydrochloride and naloxone hydrochloride) sublingual film.
- 36. On August 30, 2010, the FDA approved NDA No. 22-410 for the manufacture, marketing, and sale of Suboxone[®] sublingual film for the treatment of opioid dependence.

 Plaintiff Indivior has sold Suboxone[®] sublingual film under NDA No. 22-410 since its approval.
- 37. The '221 patent is listed in the FDA's *Approved Drug Products with Therapeutic Equivalence Evaluations* (the "Orange Book") as covering Suboxone® sublingual film.

DEFENDANTS' INFRINGING GENERIC PRODUCT

38. Defendants submitted ANDA No. 205854 to FDA under 21 U.S.C. § 355(b)(2), seeking approval to engage in commercial manufacture, use, and/or sale of Defendants' generic product before expiration of the patent-in-suit.

- 39. ANDA No. 205854 refers to and relies on Plaintiffs' NDA for Suboxone® sublingual film and purports to contain data showing bioequivalence of Defendants' generic product with Suboxone® sublingual film.
- 40. On information and belief, Defendants' generic product includes an anti-tacking agent recited in the claims of the '221 patent as part of commercially sourced polyethylene oxide.

THE PENDING ANDA LITIGATION BETWEEN THE PARTIES

- 41. Plaintiffs Indivior and Indivior UK and Defendants are involved in ongoing litigation in this District, Civil Action No. 17-7997.
- 42. Civil Action No. 17-7997 relates to Defendant Par's submission of ANDA No. 205854 to FDA seeking approval to engage in commercial manufacture, use, and/or sale of Plaintiffs' NDA for Suboxone® sublingual film.
- 43. The patent at issue in Civil Action No. 17-7997 includes U.S. Patent. No. 9,687,454 ("the '454 patent").

COUNT 1 Infringement of the '221 Patent Under 35 U.S.C. § 271(e)(2)

- 44. On information and belief, Defendants' generic product is covered by one or more claims of the '221 patent.
- 45. By filing ANDA No. 205854 under 21 U.S.C. § 355(j) for the purposes of obtaining approval to engage in the commercial manufacture, use, and/or sale of Defendants' generic product prior to the expiration of the '221 patent, Defendants have committed an act of infringement of the '221 patent under 35 U.S.C. § 271(e)(2).
- 46. Plaintiffs are entitled to the relief provided by 35 U.S.C. § 271(e)(4), including, inter alia, an order of this Court that the FDA set the effective date of approval for ANDA No.

205854 to be a date which is not any earlier than the expiration date of the '221 patent, including any extensions of that date.

COUNT 2 Declaratory Judgment of Infringement of the '221 Patent Under 35 U.S.C. § 271

- 47. On information and belief, unless enjoined by this Court, Defendants plan and intends to engage in the manufacture, use, offer for sale, sale, marketing, distribution, and/or importation of Defendants' generic product immediately following approval of ANDA No. 205854.
- 48. On information and belief, Defendants' commercial manufacture of Defendants' generic product before the expiration of the '221 patent would infringe one or more claims of the '221 patent under 35 U.S.C. § 271.
- 49. The acts of infringement by Defendants set forth above will cause Plaintiffs irreparable harm for which they have no adequate remedy at law, and those acts will continue unless enjoined by this Court.

PRAYER FOR RELIEF

WHEREFORE, Plaintiffs respectfully request that this Court enter:

- A. A Judgment that Defendants have infringed the '221 patent under 35 U.S.C. § 271(e)(2) by submitting and maintaining ANDA No. 205854;
- B. A Declaratory Judgment that Defendants' commercial manufacture within the United States of Defendants' generic product would infringe the '221 patent under 35 U.S.C. § 271;
- C. Preliminary and permanent injunctions, restraining and enjoining Defendants, its officers, agents, attorneys, affiliates, divisions, successors and employees, and those acting in privity or concert with them, from engaging in, causing, or inducing the commercial

manufacture, use, offer to sell, or sale within the United States, or importation into the United States, of drugs and formulations, or from inducing and/or encouraging the use of methods, claimed in the patent-in-suit;

- D. An Order that the effective date of any approval of ANDA No. 205854 be a date that is not earlier than the expiration of the patent-in-suit, including any extensions thereof and any later expiration of exclusivity associated with the '221 patent;
- E. A Judgment and Order finding that this is an exceptional case within the meaning of 35 U.S.C. § 285 and awarding to Plaintiffs their reasonable attorneys' fees;
- F. A Judgment granting Plaintiffs compensatory damages in an amount to be determined at trial including both pre-judgment and post-judgment interest if Defendants commercially manufacture, use, offer to sell, or sell in the United States, or import into the United States, Defendants' generic product before the expiration of the patent-in-suit, including any extensions; and
 - G. Any and all other relief as the Court deems just and proper.

Dated: February 7, 2018

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CERTIFICATION PURSUANT TO LOCAL CIVIL RULES 11.2 & 40.1

Pursuant to Local Civil Rules 11.2 and 40.1, I hereby certify that the matters captioned, Indivior Inc., et al. v. Alvogen Pine Brook, Inc., Civil action No. 17-7106 (KM)(CLW), Indivior Inc., et al. v. Dr. Reddy's Laboratories S.A., et al., Civil action No. 17-7111 (KM)(CLW), Indivior Inc., et al. v. Teva Pharmaceuticals USA, Inc., Civil action No. 17-7115 (KM)(CLW), Indivior Inc., et al. v. Par Pharmaceuticals, Inc., et al., Civil action No. 17-7997 (KM)(CLW), Indivior Inc. v. Actavis Laboratories UT, Inc., Civil Action No. 2:17-cv-01034 (D. Utah), Indivior Inc. v. Actavis Laboratories UT, Inc., Civil Action No. 2:18-cv-00124 (D. Utah), and Par Pharmaceutical Inc. v. Indivior Inc., et al., Civil Action No. 1:2017-cv-01280 (E.D.Va.) are related to the matter in controversy because they involve some of the same Plaintiffs and the same patents.

I further certify that, to the best of my knowledge, the matter in controversy is not the subject of any other action pending in any court, or of any pending arbitration or administrative proceeding.

Dated: February 7, 2018

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EXHIBIT A

(12) United States Patent

Myers et al.

(54) UNIFORM FILMS FOR RAPID-DISSOLVE DOSAGE FORM INCORPORATING ANTI-TACKING COMPOSITIONS

(71) Applicant: MonoSol Rx, LLC, Warren, NJ (US)

(72) Inventors: Garry L. Myers, Kingsport, TN (US); Pradeep Sanghvi, North Brunswick, NJ

(US); Andrew Philip Verrall, Indianapolis, IN (US); Vimala Francis, Fremont, CA (US); Laura Brooks,

Sheboygan, WI (US)

(73) Assignee: MonoSol Rx, LLC, Warren, NJ (US)

(*) Notice: Subject to any disclaimer, the term of this

patent is extended or adjusted under 35 U.S.C. 154(b) by 0 days.

This patent is subject to a terminal dis-

Appl. No.: 15/438,406

(22) Filed: Feb. 21, 2017

Prior Publication Data (65)

> US 2017/0189346 A1 Jul. 6, 2017

Related U.S. Application Data

- (63) Continuation of application No. 15/398,398, filed on Jan. 4, 2017, now abandoned, which is a continuation (Continued)
- (51) Int. Cl. A61F 13/00 (2006.01)A61K 9/70 (2006.01)

(Continued)

(52) U.S. Cl. CPC A61K 9/7007 (2013.01); A23L 33/10 (2016.08); A23L 33/105 (2016.08); A23L *33/15* (2016.08);

(Continued)

US 9,855,221 B2 (10) **Patent No.:**

(45) Date of Patent:

*Jan. 2, 2018

CPC A61K 9/006 See application file for complete search history.

(56)References Cited

(58) Field of Classification Search

U.S. PATENT DOCUMENTS

26,401 A 12/1859 Brashear et al. 307,537 A 11/1884 Foulks (Continued)

FOREIGN PATENT DOCUMENTS

AU741362 B2 11/2001 2274910 C 7/2005 CA(Continued)

OTHER PUBLICATIONS

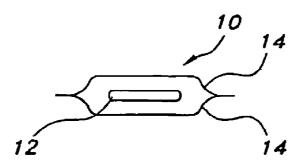
"Adsorption at Solid Surfaces," Encyclopedia of Pharmaceutical Technology (Swarbrick (ed.)), pp. 73 (1988). (Continued)

Primary Examiner — Benjamin Packard (74) Attorney, Agent, or Firm — Hoffmann & Baron, LLP

(57)ABSTRACT

The present invention relates to water-soluble films incorporating anti-tacking agents and methods of their preparation. Anti-tacking agents may improve the flow characteristics of the compositions and thereby reduce the problem of film adhering to a user's mouth or to other units of film. In particular, the present invention relates to edible watersoluble delivery systems in the form of a film composition including a water-soluble polymer, an active component selected from cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof and at least one anti-tacking agent.

30 Claims, 3 Drawing Sheets



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11/1970 Oberhofer Related U.S. Application Data 3,539,605 A 3,551,556 A 12/1970 Kliment et al. of application No. 15/144,191, filed on May 2, 2016, 3,598,122 A 8/1971 Zaffaroni 3,610,248 A 10/1971 now abandoned, which is a continuation of applica-Davidson 3,625,351 A 12/1971 Eisenberg tion No. 14/844,810, filed on Sep. 3, 2015, now 3,632,740 A 1/1972 Robinson et al. abandoned, which is a continuation of application No. 3,640,741 A 2/1972 Etes 14/284,019, filed on May 21, 2014, now abandoned, 2/1972 3.641.237 A Gould et al. which is a continuation of application No. 11/517, 3,650,461 A 3/1972 Hutcheson 982, filed on Sep. 8, 2006, now Pat. No. 8,765,167, 3,677,866 A 7/1972Pickett et al. 3,731,683 A 5/1973 Zaffaroni which is a continuation-in-part of application No. 3,753,732 A 8/1973 Boroshok 10/074,272, filed on Feb. 14, 2002, now Pat. No. 3,755,558 A 8/1973 Scribner 7,425,292. 3,768,725 A 10/1973 Pilaro 3,795,527 3/1974 Stone et al. (60) Provisional application No. 60/715,528, filed on Sep. 3,797,494 A 3/1974 Zaffaroni 9, 2005, provisional application No. 60/328,868, filed 3,809,220 A 5/1974 Arcudi 6/1974 on Oct. 12, 2001. 3,814,095 A Lubens 3,825,014 A 7/1974 Wroten 3,835,995 A 9/1974 Haines (51) Int. Cl. 3,840,657 10/1974 Nortleet A61K 9/00 (2006.01)3,892,905 A 7/1975 Albert A61K 47/12 (2006.01)3,911,099 A 10/1975 DeFoney et al. 3.933.245 A 1/1976 Mullen A61K 47/02 (2006.01)3,972,995 A 8/1976 Tsuk et al. A61K 47/22 (2006.01)3,979,839 A 9/1976 Blanie A61K 47/34 (2017.01)3,996,934 A 12/1976 Zaffaroni A61K 47/38 (2006.01)3,998,215 12/1976 Anderson et al. A61K 47/36 (2006.01)4,015,023 A 3/1977 Lamberti et al. 4,022,924 A 5/1977 Mitchell et al. A61K 31/485 (2006.01)4,029,757 A 6/1977 Mlodozeniec et al A23L 33/16 (2016.01)4.029.758 A 6/1977 Mlodozeniec et al. A23L 33/15 (2016.01)4.031.200 A 6/1977 Reif A23L 33/105 (2016.01)9/1977 4,049,848 A Goodale et al. 4,053,046 A 10/1977 A23L 33/10 (2016.01)Roark 4,067,116 A 1/1978 Bryner et al. A23P 20/20 (2016.01)4,105,116 A 8/1978 Jones et al. (52)U.S. Cl. 4,123,592 A 10/1978 Rainer et al. CPC 4,126,503 A 11/1978 Gardner (2016.08); A61K 9/0056 (2013.01); A61K 4.128.445 A 12/1978 Sturzenegger et al. 1/1979 4,136,145 A 31/485 (2013.01); A61K 47/02 (2013.01); Fuchs et al 1/1979 4.136.162 A Fuchs et al A61K 47/12 (2013.01); A61K 47/22 (2013.01); 2/1979 4,139,627 A Lane et al. **A61K** 47/34 (2013.01); **A61K** 47/36 (2013.01); 4,202,966 A 5/1980 Misaki et al A61K 47/38 (2013.01); A23V 2002/00 4,226,848 A 10/1980 Nagai et al. 4,249,531 A 2/1981 (2013.01)Heller et al. 4,251,400 A 2/1981 Columbus 4.251.561 A 2/1981 Gajewski (56)References Cited 4,284,194 A 8/1981 Flatau 4,284,534 A 8/1981 Ehrlich U.S. PATENT DOCUMENTS 4.292.299 9/1981 Suzuki et al. 4,294,820 A 10/1981 Keith et al. 476,085 A 5/1892 Smith 4,302,465 A 11/1981 Af Ekenstam et al. 492,417 A 2/1893 McAlister 4,307,075 A 12/1981 Martin 503,070 A 8/1893 Broadwell et al. 4,307,117 A 12/1981 Leshik 596,302 A 12/1897 McMahon 4,325,855 A 4/1982 Dickmann et al. 688,446 A 10/1901 Stempel, Jr. 4,341,563 A 7/1982 Kurihara et al. 1,110,546 A 9/1914 Hewitt 4,365,423 A 12/1982 Arter et al. 1,827,354 A 10/1931 Cooper 4,373,036 A 2/1983 Chang et al. 2,142,537 A 1/1939 Tiaxa 4,390,450 A 6/1983 Gibson et al 3/1942 2,277,038 A Curtis 4,406,708 A 9/1983 Hesselgren 7/1944 2,352,691 A Curtis 4,432,975 A 2/1984 Libby 2,376,656 A 5/1945 Leonia 4,438,258 A 3/1984 Graham 2,501,544 A 3/1950 Shrontz 4,451,260 A 5/1984 Mitra 2,612,165 A 9/1952Szukerski 4,460,532 A 7/1984 Cornell 2,980,554 A 4/1961 Gentile et al. 4,460,562 A 7/1984 Keith et al. 3,007,848 A 11/1961 Stroop 4.466,973 A 8/1984 Rennie 3.044.338 A 7/1962 Horton et al. 4,478,658 A 10/1984 Wittwer 3,131,068 A 4/1964 Grief 4,483,846 A 11/1984 Koide et al. 3,142,217 A 7/1964 Busse 3/1985 Eby, III 4,503,070 A 3,189,174 A 6/1965 Cormack 4.511,592 A 4/1985 Percel et al. 3,237,596 A 3/1966 Grass, Jr. et al. 4.515.162 A 5/1985 Yamamoto et al. 3,242,959 A 3/1966 Glass 4,517,173 A 5/1985 Kizawa et al. 3,249,109 A 5/1966 Maeth et al. 4,529,301 A 7/1985 Rountree 6/1967 3,324,754 A Peavy 4,529,601 A 7/1985 Broberg et al. 3.370.497 A 2/1968 Busse

4,529,748 A

4,562,020 A

4,568,535 A

4,569,837 A

7/1985

2/1986 Loesche

2/1986 Suzuki et al

12/1985

Wienecke

Hijiya et al.

3,419,137 A

3,444,858 A

3,451,539 A

3,536,809 A

12/1968

5/1969

6/1969

Walck, III

Russell

10/1970 Applezwig

Wysocki

US 9,855,221 B2 Page 3

(56)	Referen	ces Cited	5,028,632		7/1991	
IJ.	S. PATENT	DOCUMENTS	5,044,241 5,044,761			Labrecque Yuhki et al.
0.	5. 111121.11	DOCOMENTO	5,045,445			Schultz
4,572,832 A		Kigasawa et al.	5,047,244 5,049,322			Sanvordeker et al. Devissaguet et al.
4,582,835 A 4,585,452 A		Lewis et al. Sablotsky	5,056,584		10/1991	
4,588,592 A			5,064,717	A	11/1991	Suzuki et al.
4,593,053 A	6/1986	Jevne et al.	5,072,842		12/1991	
4,598,089 A		Hadvary et al.	5,078,734 5,089,307		1/1992 2/1992	Ninomiya et al.
4,608,249 A 4,613,497 A		Otsuka et al. Chavkin	5,100,591			Leclef et al.
4,615,697 A	10/1986	Robinson	5,107,734			Armbruster Hirashima
4,619,701 A		Angrick et al.	5,116,140 5,118,508			Kikuchi et al.
4,621,482 A 4,623,394 A		Crevasse et al. Nakamura et al.	5,126,160			Giddey et al.
4,631,837 A	12/1986	Magoon	5,137,729			Kuroya et al.
4,639,367 A		Mackles	5,158,825 5,166,233			Altwirth Kuroya et al.
4,648,509 A 4,659,714 A		Watt-Smith	5,176,705		1/1993	
4,661,359 A		Seaborne et al.	5,184,771			Jud et al.
4,675,009 A		Hymes et al.	5,186,938 5,188,838		2/1993	Sablotsky et al. Deleuil et al.
4,695,465 A 4,704,119 A		Kigasawa et al. Shaw et al.	5,196,436		3/1993	
4,705,174 A			5,229,164			Pins et al.
4,712,460 A		Allen et al.	5,230,441 5,234,957			Kaufman et al. Mantelle
4,713,239 A 4,713,243 A		Babaian et al. Schiraldi et al.	5,264,024			Bosvot et al.
4,713,251 A		Seighman	5,271,940	A	12/1993	Cleary et al.
4,716,802 A	1/1988	O'Connor et al.	5,272,191 5,273,758		12/1993 12/1993	Ibrahim et al.
4,722,761 A 4,727,064 A		Cartmell et al.	5,293,699			Faust et al.
4,740,365 A		Yukimatsu et al.	5,316,717	A	5/1994	Koepff et al.
4,748,022 A	5/1988	Busciglio	5,325,968			Sowden
4,752,465 A		Mackles	5,328,942 5,344,676			Akhtar et al. Kim et al.
4,762,230 A 4,764,378 A		Keith et al.	5,346,701			Heiber et al.
4,765,983 A	8/1988	Takayanagi et al.	5,354,551			Schmidt
4,772,470 A		Inoue et al.	5,360,629 5,369,131			Milbourn et al. Poli et al.
4,777,046 A 4,780,309 A		Iwakura et al. Geria et al.	5,375,930		12/1994	Tani
4,781,294 A			5,380,529			Heusser et al.
4,787,517 A			5,393,528 5,405,637		2/1995 4/1995	Staab Martinez et al.
4,789,667 A 4,802,924 A		Makino et al. Woznicki et al.	5,407,278		4/1995	
4,828,841 A		Porter et al.	5,411,945			Ozaki et al.
4,849,246 A		Schmidt	5,413,792 5,422,127			Ninomiya et al. Dube et al.
4,851,394 A 4,860,754 A		Kubodera Sharik et al.	5,423,423		6/1995	Sato et al.
4,861,632 A		Caggiano	5,433,960			Meyers
RE33,093 E		Schiraldi et al.	5,451,419 5,455,043		9/1995	Schwab et al. Fischel-Ghodsian
4,872,270 A 4.876.092 A		Fronheiser et al. Mizobuchi et al.	5,458,884			Britton et al.
4,876,970 A			5,462,749	A	10/1995	
4,880,416 A		Horiuchi et al.	5,472,704 5,479,408		12/1995 12/1995	Santus et al.
4,888,354 A 4,894,232 A		Chang et al. Reul et al.	5,489,436			Hoy et al.
4,900,552 A		Sanvordeker et al.	5,506,046			Andersen et al.
4,900,554 A		Yangibashi et al.	5,506,049 5,518,902			Swei et al. Ozaki et al.
4,900,556 A 4,910,247 A		Wheatley et al. Haldar et al.	5,529,782		6/1996	
4,915,950 A		Miranda et al.	5,530,861			Diamant et al.
4,925,670 A		Schmidt	5,550,178 5,551,033			Desai et al. Foster et al.
4,927,634 A 4,927,636 A		Sorrentino et al. Hijiya et al.	5,552,152		9/1996	
4,929,447 A			5,553,835			Dresie et al.
4,937,078 A		Mezei et al.	5,560,538 5,567,237			Sato et al. Kapp-Schwoerer et al.
4,940,587 A 4,948,580 A		Jenkins et al. Browning	5,567,431			Vert et al.
4,958,580 A		Asaba et al.	5,573,783	A	11/1996	Desieno et al.
4,978,531 A	12/1990	Yamazaki et al.	5,582,342		12/1996	
4,980,169 A 4,981,693 A		Oppenheimer et al. Higashi et al.	5,587,175 5,588,009		12/1996 12/1996	Viegas et al.
4,981,875 A		Leusner et al.	5,589,357	Α		Martinez et al.
4,993,586 A	2/1991	Taulbee et al.	5,593,697	A	1/1997	Barr et al.
5,023,082 A		Friedman et al.	5,595,980			Brode et al.
5,023,271 A 5,024,701 A		Vigne et al. Desmarais	5,601,605 5,605,696			Crowe et al. Eury et al.
5,025,692 A		Reynolds	5,605,698		2/1997	
,,	· -	•	· / ·			

US 9,855,221 B2 Page 4

(56)		Referen	ces Cited	6,072,100			Mooney et al.
	211	PATENT	DOCUMENTS	6,074,097 6,077,558		6/2000	Hayashi et al. Euber
	0.5.	TILIVI	DOCUMENTS	6,090,401			Gowan, Jr. et al.
	5,613,779 A	3/1997	Niwa	6,099,871			Martinez
	5,614,212 A		D'Angelo et al.	6,103,266			Tapolsky et al.
	5,620,757 A		Ninomiya et al.	6,106,930 6,143,276		11/2000	Ludwig Unger
	5,629,003 A 5,629,021 A	5/1997 5/1997	Horstmann et al.	6,148,708		11/2000	
	5,633,006 A		Catania et al.	6,152,007		11/2000	
	5,641,093 A		Dolin et al.	6,153,210			Roberts et al.
	5,641,536 A		Lech et al.	6,153,220			Cumming et al.
	D380,836 S		Fitzpatrick et al.	6,159,498 6,161,129			Tapolsky et al. Rochkind
	5,647,431 A 5,653,993 A		Takeshita et al. Ghanta et al.	6,177,066			Pataut et al.
	5,656,296 A		Khan et al.	6,177,092		1/2001	Lentini et al.
	5,656,297 A		Bernstein et al.	6,177,096			Zerbe et al.
	5,670,168 A		Baichwal et al.	6,183,808 6,197,329			Grillo et al. Hermelin et al.
	5,679,145 A		Andersen et al.	6,203,566			Alanen et al.
	5,681,873 A 5,689,550 A		Norton et al. Garson et al.	6,219,694			Lazaridis et al.
	5,698,181 A	12/1997		6,221,402			Itoh et al.
	5,698,217 A	12/1997	Wilking	6,227,359			Truluck
	5,700,478 A		Biegajski et al.	6,230,894 6,231,957			Danville Zerbe et al.
	5,700,479 A 5,725,648 A		Lundgren Brown et al.	6,238,700			Dohner et al.
	5,733,575 A		Mehra et al.	6,264,981			Zhang et al.
	5,738,211 A	4/1998	Ichino et al.	6,267,808			Grillo et al.
	5,742,905 A		Pepe et al.	6,268,048			Topolkaraev et al.
	5,750,145 A	5/1998		6,284,264 6,287,595			Zerbe et al. Loewy et al.
	5,750,157 A 5,750,585 A		Grosswald et al. Park et al.	6,294,206			Barrett-Reis et al.
	5,759,599 A		Wampler et al.	6,311,627	B1	11/2001	Draper et al.
	5,761,525 A		Williams	6,338,407			Danville
	5,764,639 A		Staples et al.	6,344,088 6,374,715			Kamikihara et al. Takatsuka
	5,764,899 A		Eggleston et al. Foster et al.	6,375,963			Repka et al.
	5,765,004 A 5,766,332 A		Graves et al.	6,391,294			Dettmar et al.
	5,766,525 A		Andersen et al.	6,394,306			Pawlo et al.
	5,766,620 A		Heiber et al.	6,395,299			Babich et al.
	5,766,839 A		Johnson et al.	6,413,792 6,419,903			Sauer et al. Xu et al.
	5,771,353 A 5,785,180 A		Eggleston et al. Dressel et al.	6,419,906			Xu et al.
	5,792,494 A		Kanca et al.	6,428,825			Sharma et al.
	5,800,832 A		Tapolsky et al.	6,432,460			Zietlow et al.
	5,806,284 A		Gifford	6,436,464 6,454,788		8/2002 9/2002	
	5,815,398 A		Dighe et al. Waskiewicz	6,467,621		10/2002	
	5,822,526 A 5,830,437 A		Ascione et al.	6,468,516			Geria et al.
	5,830,884 A		Kasica et al.	6,472,003			Barrett-Reis et al.
	5,846,557 A		Eisenstadt et al.	6,482,517			Anderson
	5,847,023 A		Viegas et al.	6,488,963 6,495,599			McGinity et al. Auestad et al.
	5,862,915 A 5,864,684 A	1/1999	Plezia et al. Nielsen	6,503,532			Murty et al.
	5,881,476 A		Strobush et al.	6,509,072	B2		Bening et al.
	5,891,461 A	4/1999	Jona et al.	6,534,090			Puthli et al.
	5,891,845 A	4/1999	Myers	6,534,092 6,552,024		3/2003	Chen et al.
	5,894,930 A 5,900,247 A		Faughey et al. Rault et al.	6,562,375		5/2003	
	5,906,742 A		Wang et al.	6,575,999		6/2003	Rohrig
	5,930,914 A	8/1999	Johansson et al.	6,589,576			Borschel et al.
	5,937,161 A	8/1999	Mulligan et al.	6,592,887 6,596,298			Zerbe et al. Leung et al.
	5,941,393 A	8/1999	Wilfong, Jr. Chorosinski et al.	6,596,302			O'Connor et al.
	5,945,651 A 5,948,430 A		Zerbe et al.	6,599,542			Abdel-Malik et al.
	5,955,097 A		Tapolsky et al.	6,610,338			Tang
	5,965,154 A	10/1999	Haralambopoulos	6,620,440			Hsia et al.
	5,980,554 A		Lenker et al.	6,655,112 6,656,493			Cremer et al. Dzija et al.
	5,992,742 A 5,995,597 A	11/1999	Sullivan et al. Woltz et al.	6,660,292			Zerbe et al.
	6,004,996 A	12/1999		6,667,060			Vandecruys et al.
	6,024,975 A		D'Angelo et al.	6,668,839	B2	12/2003	Williams
	6,030,616 A	2/2000	Waters et al.	6,708,826			Ginsberg et al.
	6,031,895 A		Cohn et al.	6,709,671		3/2004	
	6,036,016 A	3/2000		6,720,006			Hanke et al.
	6,047,484 A 6,051,253 A		Bolland et al. Zettler et al.	6,726,054 6,730,319			Fagen et al. Maeder et al.
	6,054,119 A		Hurme et al.	6,752,824		6/2004	
	6,064,990 A		Goldsmith	6,776,157			Williams et al.
	•						

US 9,855,221 B2 Page 5

(56)	Referen	ces Cited	2004/0241242 2005/0003048			Fuisz et al. Pearce et al.
U.S	S. PATENT	DOCUMENTS	2005/0011776		1/2005	
			2005/0019588			Berry et al.
6,797,283 B1		Edgren et al.	2005/0035133 2005/0037055		2/2005 2/2005	Gerulski et al. Yang et al.
6,800,329 B2 6,824,829 B2		Horstmann et al. Berry et al.	2005/0048102		3/2005	Tang et al. Tapolsky et al.
6,865,860 B2		Arakawa et al.	2005/0055123	A1	3/2005	Franz
6,905,016 B2	6/2005	Kanios et al.	2005/0089548			Virgalitto et al.
6,913,766 B1		Krumme et al.	2005/0095272 2005/0115862		5/2005 6/2005	Augello Maietta
6,929,399 B2 6,929,400 B2		Nokura Razeti et al.	2005/0118217			Barnhart et al.
7,005,142 B2		Leon et al.	2005/0118271			Schliecker et al.
7,040,503 B2		Leichter et al.	2005/0136115 2005/0147658			Kulkarni et al. Tapolsky et al.
7,067,116 B1 7,093,736 B2		Bess et al. Maietta et al.	2005/0163714		7/2005	
7,115,507 B2	10/2006	Kawase	2005/0170138	A1	8/2005	Berry
7,179,788 B2	2/2007	DeFelippis et al.	2005/0191349			Boehm et al. Palermo et al.
7,241,411 B2		Berry et al.	2005/0192309 2005/0214251			Pohl et al.
7,357,891 B2 7,390,503 B1		Yang et al. Ahmed et al.	2005/0222781			Yue et al.
7,425,292 B2	9/2008	Yang et al.	2005/0232977			Khan et al.
7,428,859 B2		Fujita et al.	2005/0239845 2006/0023976		2/2006	Proehl et al. Alvater et al.
7,484,640 B2 7,531,191 B2		von Falkenhausen et al. Zion et al.	2006/0039958			Fuisz et al.
7,579,019 B2		Tapolsky et al.	2006/0071057		4/2006	Aschenbrenner et al.
7,591,801 B2		Brauker et al.	2006/0073190			Carroll et al.
7,665,896 B1			2006/0083786 2006/0093679			Chaudhari et al. Mayer et al.
7,666,337 B2 7,694,617 B2		Yang et al. Habra et al.	2006/0104910		5/2006	
7,824,588 B2		Yang et al.	2006/0147493			Yang et al.
7,910,031 B2		Yang et al.	2006/0163269 2006/0180604		7/2006	Anderson et al. Ginsberg et al.
8,017,150 B2 8,051,983 B2		Yang et al. Simon et al.	2006/0180004			Wu et al.
8,147,866 B2		Finn et al.	2006/0189772			Scheibel et al.
2001/0006677 A1	7/2001	McGinity et al.	2006/0198790			Dugger, III et al.
2001/0022964 A1		Leung et al.	2006/0198885 2006/0210610			Dharmadhikari et al. Davidson et al.
2001/0046511 A1 2002/0006677 A1		Zerbe et al. Egermeier et al.	2006/0213348		9/2006	
2002/0012689 A1		Stillman	2006/0215941			Golbert
2002/0045582 A1		Margolin et al.	2006/0246141 2006/0264448		11/2006	Liversidge et al. Pryde
2002/0098198 A1 2002/0104774 A1		Walls et al. Hammond	2006/0281775			Kelly, II et al.
2002/0127254 A1		Fotinos et al.	2006/0286108		12/2006	
2002/0131990 A1		Barkalow et al.	2007/0027213 2007/0045148		2/2007 3/2007	Oberegger et al. Saclier et al.
2002/0147201 A1 2002/0170567 A1		Chen et al. Rizzotto et al.	2007/0043148			Yang et al.
2002/0170307 A1 2002/0177380 A1		Forman et al.	2007/0087036			Durschlag et al.
2003/0035841 A1	2/2003	Dzija et al.	2007/0098746			Nichols et al.
2003/0044511 A1		Zerbe et al.	2007/0122455 2007/0138049		5/2007 6/2007	Myers et al. Bitner
2003/0054039 A1 2003/0068378 A1		Zyck et al. Chen et al.	2007/0148097			Finn et al.
2003/0069263 A1	4/2003	Breder et al.	2007/0170196			Libohova et al.
2003/0072865 A1		Bindels et al.	2007/0205127 2007/0231368			Barndt et al. Wang et al.
2003/0077315 A1 2003/0107149 A1		Lee et al. Yang et al.	2007/0251308			Fuisz et al.
2003/0107149 A1 2003/0118649 A1		Gao et al.	2007/0281003		12/2007	
2003/0121932 A1			2008/0044454			Yang et al.
2003/0124176 A1		Hsu et al.	2008/0073235 2008/0075825			Harada et al. Fuisz et al.
2003/0140760 A1 2003/0147956 A1		Shefer et al.	2008/0081071		4/2008	Sanghvi et al.
2003/0161926 A1	8/2003	Kemp et al.	2008/0105582			Ludwig et al.
2003/0183643 A1		Fagen et al.	2008/0233174 2008/0242558			Myers et al. Belcher et al.
2003/0224044 A1 2004/0013731 A1		Chen et al.	2008/0242736		10/2008	Fuisz
2004/0024003 A1		Asmussen et al.	2008/0254105		10/2008	Tapolsky et al.
2004/0044367 A1		,	2008/0260805 2008/0260809		10/2008 10/2008	Yang et al. Yang et al.
2004/0058457 A1 2004/0091677 A1		Huang et al. Topolkaraev	2008/0268116		10/2008	Kring
2004/0091677 A1 2004/0096569 A1		Barkalow et al.	2008/0290106	A1	11/2008	van der Klaauw et al.
2004/0102867 A1	5/2004	Palanisamy et al.	2008/0299197		12/2008	Toneguzzo et al.
2004/0111275 A1		Kroll et al.	2008/0300173			DeFrees Intini
2004/0120991 A1 2004/0136924 A1		Gardner et al. Boyd et al.	2008/0308449 2009/0004254		1/2008	Intini Maibach
2004/0130924 A1 2004/0137458 A1		Archambault et al.	2009/0009332			Nunez et al.
2004/0156901 A1	8/2004	Thakur et al.	2009/0014491	A1	1/2009	Fuisz et al.
2004/0191302 A1		Davidson	2009/0029074		1/2009	Sasine et al.
2004/0209057 A1		Enlow et al.	2009/0074333 2009/0104270		3/2009	Griebel et al. Myers et al.
2004/0219109 A1	11/2004	TIAICII	2009/01042/0	AI	4/2009	iviyers et al.

Page 6

(56)	References Cited	JP	473268 A	
		JP	5147140 A	
	U.S. PATENT DOCUMENTS	JР	7322812 A	
		JР	11255247 A	
	46336 A1 6/2009 Masi	JР	2000159658 A	
	31075 A1 7/2009 Gordon et al.	JP JP	2001048196 A 2001225851 A	
	92075 A1 7/2009 Steiner	JP JP	2001223831 A	
	96907 A1 8/2009 Bunick et al.	JР	2001279100 F	
	97614 A1 12/2009 Rademacher et al. 15128 A1 1/2010 Lee et al.	JР	2004222663 A	
	37470 A1 4/2010 Chee et al.	JР	2006143335 A	
	92545 A1 4/2010 Yang et al.	JP	2008011194 A	1/2008
	78254 A1 7/2010 Hariharan et al.	WO	1988007103	9/1988
	21309 A1 9/2010 Myers et al.	WO	9105540 A	
2010/029	97232 A1 11/2010 Myers et al.	WO	1992012704	8/1992
2011/018	89259 A1 8/2011 Vasisht et al.	WO	9215289 A	
2011/026	52522 A1 10/2011 Finn et al.	WO	9505416 A	
		WO WO	9518046 <i>A</i> 1995023596	A1 7/1995 9/1995
	FOREIGN PATENT DOCUMENTS	WO	9530601 A	
		WO	9615903 A	
CA	2317491 C 6/2008	wo	9625150 A	
CH	639619 A5 11/1983	WO	1996025638	8/1996
CN	1118254 A 3/1996	WO	9731621 A	1 9/1997
DE	2746414 A1 4/1979	WO	9732573 A	1 9/1997
DE	2449865 B2 6/1981	WO	1997044016	11/1997
DE DE	2432925 C3 11/1985 3630603 C2 6/1989	WO	9810993 A	
DE DE	19646392 A1 5/1998	WO	9817251 A	
DE	202004003781 U1 5/2004	WO	1998014179	4/1998
EP	0014253 A2 8/1980	WO	9935051 A	
EP	0021178 B1 1/1981	WO WO	955312 A 200002536	11/1999 1/2000
EP	0090560 A2 10/1983	WO	200002330	1/2000
EP	0095892 A1 12/1983	wo	0018365 A	
EP	0065370 B1 1/1985	WO	2000/027618 A	
EP	0248548 B1 5/1987	WO	0024647 A	
EP	0232877 A2 8/1987	WO	0042992 A	
EP	0241178 10/1987	WO	2000057858	10/2000
EP EP	0285568 A2 3/1988	WO	2001003917 A	
EP EP	0274431 A2 7/1988 0219762 B1 12/1990	WO	0130288 A	
EP	0259749 B1 8/1991	WO	2001034121	5/2001
EP	0200508 B1 10/1991	WO	0143728 A	
EP	0241178 B1 1/1992	WO WO	0156904 A 0168452 A	
EP	0514691 A2 4/1992	WO	0170194 A	
EP	0273069 B1 10/1992	wo	0170194 A	
EP	0250187 B1 9/1993	WO	0191721 A	
EP	0452446 B1 12/1993	WO	0205789 A	
EP	0598606 A1 5/1994	WO	0207711 A	1/2002
EP	0381194 B1 8/1994	WO	2002005820 A	
EP EP	0440462 B1 12/1994 0636364 A1 1/1995	WO	2006017462 A	
EP	0450141 B1 5/1995	WO	0243657 A	
EP	0460588 B1 8/1995	WO	2002/064148 A	
EP	0514691 B1 1/1996	WO	02062315 A 02074238 A	
EP	0598606 B1 6/1999	WO WO	02074238 A 02091965 A	
EP	1143940 7/2000	wo	03011259 A	
EP	1110546 A1 6/2001	WO	03015749 A	
EP	1177788 A2 2/2002	WO	03030881 A	
EP	1219291 A1 3/2002	WO	03030882 A	
EP	1243523 A1 9/2002	WO	03030883 A	4/2003
EP	0949925 B1 1/2004	WO	03043659 A	
EP EP	1504765 A1 2/2005 1267829 B1 5/2006	WO	2003/101357 A	
EP	1674078 A2 6/2006	WO	2004009445 A	
EP	1852041 A2 11/2007	WO	2004035407 A	
EP	1897543 A1 3/2008	WO WO	2004043165 A 2004045305 A	
EP	1591106 B1 7/2009	WO	2004045537 A	
EP	2105389 A1 9/2009	WO	2004052335 A	
EP	2253224 A1 11/2010	wo	2004060298 A	
EP	2305310 A1 4/2011	WO	2004087084 A	
FR	2716098 A1 8/1995	WO	2004113193 A	
GB	1061557 3/1967	WO	2005020933 A	
GB GB	1154317 6/1969 1510000 5/1078	WO	2005035776 A	
GB GB	1510999 5/1978 2166651 A 5/1986	WO	2005039499 A	
GB GB	2447016 A 9/2009	WO	2005074867 A	
JP	56100714 A 8/1981	WO	2005102287 A	
ĴР	62126950 A 6/1987	WO	2005102863 A	11/2005
JP	2265444 A 10/1990	WO	2005123074 A	12/2005

Page 7

(56)	References Cited					
	FOREIGN PATENT DOCUMENTS					
WO W	2006004480 A1 1/2006 2006031209 A1 3/2006 2006037979 A2 4/2006 2006039264 A1 4/2006 2006037425 A1 8/2006 2006085210 A1 8/2006 2006133948 A2 12/2006 2007015105 A2 2/2007 2007067494 A1 6/2007 2007070632 A2 6/2007 2008011194 A2 1/2008 2008025791 A1 3/2008 2008036299 A2 3/2008 2008040534 A2 4/2008					
WO WO WO	2009044118 A2					

OTHER PUBLICATIONS

Photograph of Tetracycline HCL (https://de.wikipedia.org/wiki/Tetracycline#/media/File:Tetracycline-HCL_substance_photo.jpg). Textbook of Polymer Science (2nd Ed.) pp. 1-22 (Wiley 1971).

Thimmashetty, J. et al, "Preparation and Evaluation of Buccal Dosage Forms of Insulin.".

Thimmashetty, J. et al, "Design and In Vivo Evaluation of Carvedilol Buccal Mucoadhesive Patches," Pak. J. Pharm. Sci. 21(3):241-248 (2008).

Elemente des Apparatebause, (Titz, H. (ed.)), pp. 546-669 (Springer-Verlag 1992). (includes partial English translation.).

The United States Pharmacopeia (20th Rev.), pp. 3-4, 12, 16, 955-957, 1023, 1030-1031, 1412, 1451 (USP 1980).

Varanda, F. et al., "Solubility of Antibiotics in Different Solvents. 1. Hydrochloride Forms of Tetracycline, Moxifloxacin, and Ciprofloxacin," Ind. Eng. Chem. Res. 45:6368-6374 (2006).

Phramazeutische Technologie fur Studium and Beruf (Voigt, R. (ed.)), p. 65 (Ullstein Mosby 1995).

Polymer Molecular Weights (Slade, P.E. (ed.), p. 1-8 (Marcel Dekker, Inc. 1975).

Metallic Pigments in Polymers, p. 132 (Rapra Technology Limited 1999)

White, J.G., "In Situ Determination of Delavirdine Mesylate Particle Size in Solid Oral Dosage Forms," Pharmaecutical Research 16(4):545-548 (1999).

Yamamura, K. et al., "Oral Mucosal Adhesive Film Containing Local Anesthetics: In Vitro and Clinical Evaluation," J. Biomed. Mater. Res. (Appl. Biomater) 43:313-317 (1998).

Pharmazeutische Technologie: Insustrielle Herstellung und Entwicklung von Arzneimitteln (Zimmermann, I. (ed.)), p. 246 (Springer-Verlag 1998).

Modern coating technology systems for paper, film and foil (Shepherd, R (ed.)), p. 5 (Emap Maclaren Ltd. 1995).

Blank, Z. et al., "Structural studies of organic gels by SEM", J. Material Science 9:1815-1822 (1974).

CAS Presents, "Common Chemistry", http://www.commonchemistry.org.ChemicalDetail.aspx?ref=25322-68-3 &terms=polyeth . . . Oct. 28, 2009.

Huus et al., "Thermal Dissociation and Unfolding of Insulin", Biochemistry, 44: 11171-11177 (2005).

Steiner et al., "Organic Derivatives of Alginic Acid", Industrial and Engineering Chemistry, 43(9): 2073-2077 (1951).

"Cellulose" Kirk-Othmer Concise Encyclopeida of Chemical Technology, Abridged version of the 24 Volume, NY, Wiley; 227-228 (1978-1984).

"Excipients, Croscarmellose Sodium", Pformulate Excipients, http://www.pformulate.com/croscarmellose.htm (Sep. 29, 2002). Atridox(R) (Doxycycline Hyclate) Product Label.

Barton, S. et al "Citric Buffer Calculation", Version 1.1, Nov. 19,

Birkhauser, "Cell Encapsulation Technology and Therapeutics" (Jan. 5, 2009).

Bodmeier, Roland, "Evaluation of Drug-Containing Polymer Films Prepared from Aqueous Latexes", Pharmaceutical Research, vol. 6, No. 8 (1989).

Cholewinski et al., Pharmaceutica Acta Helvetiae, 71:405-419,

Croscarmellose sodium http://www.nbent.com/crosscarmellose.htm (Mar. 29, 2005).

Delsym Product Label (Feb. 13, 2007).

Di Donato et al., J. Biol. Chem, 268(7): 4745-4751, 1993.

Eiamtrakarn et al., "Gastrointestinal Mucoadhesive Path System (GI-MAPS) for oral administration of G-CSF, a model protein", Bipmaterials 23: 145-152 (2002).

Endo and Ueda, Fabad J. Pharm. Sci., 29:27-38, 2004.

Engel, June V PhD, "The Benefits of Eating Fibre" http://www.diabetes.ca/common/PrintVersion.asp?ID=45493 May 11, 2005.

Flick, E., Water-Soluble Resins—An Industrial Guide, 1991 (2nd Ed.) William Andrew Publishing/Noyes, pp. 389-392.

Goldberg et al., "Biotechnology and Food Ingredients", Springer: 352 (1991).

Hadvary et al., "Inhibition of pancreatic lipase in vitro by the covalent inhibitor tetrahydrolistatin", Biochem J.; 256: 357-361 (1988)

Ko et al., "Behavior of etrahydrolipstatin in biological model membranes and emulsions", J. of Lipid Research; 38:1544-1552 (1997).

Kuhtreiber. In Cell Encapsulation and Therapeutics . Copyright

Lazaridou et al.; Thermophysical properties of chitosan, chitosanstarch and chitosan-pullulan films near the glass transition; Elsevier Science Ltd.; 2002; pp. 179-190.

Leathers, Appl. Microbiol. Biotechnol., 62: 468-473, 2003.

Le Person, S. Le et al., "Near infrared drying of pharmaceutical thin films: experimental analysis of internal mass transport," Chemical Engineering and Processing; (1998) pp. 257-263, 37.

Mahmood et al., "A limited sampling method for the estimation of AUC and Cmax of carbamazepine and carbamazepine epoxide following a single and multiple dose of a sustained-release product", BR J Clin Pharmacol; 45:241-246 (1998).

Mix. http://www.askoxford.com/concise_oed/mixx?view=uk. Accessed Dec. 23, 2004.

Nicorete Packaging (Aug. 29, 2006).

Oriski, S.C., "Johnson debuts cutter for new Saran film" Packaging World Oct. 1, 2004, http://www.packworld.com/view-18051.

Peh Kok Khiang et al., "Polymeric Films as Vehicle for Buccal Delivery: Swelling, Mechanical, and Bioadhesive Properties," J Pharm Pharmaceut Sci (1999) pp. 53-61, 2:2.

Polyethylenglykoke, Fachgebit Chemie, Unterthema Makromolekulare Chemie, XP-002298105 (Sep. 20, 2004).

Repka et al., "Bioadhesive properties of hydroxypropylcellulose topical films produced by hot-melt extrusion," Journal of Controlled Release, 70: 341-351 (2001).

Repka et al., "Influence of Vitamin E TPGS on the properties of hydrophilic films produced by hot-melt extrusion," Int. J. Pharmaceutics, 202: 63-70 (2000).

Senel, S., et al., "Chitosan films and hydrogels of chlorhexidine gluconate for oral mucosal delivery", Int. J. Pharmaceutics, 193: pp. 197-203 (2000).

Stella, V., et al., "Gliadin Films. I: Preparation and in vitro evaluation as a carrier for controlled drug release", Int., J. Pharmaceutics, 121: pp. 117-121 (1995).

Sudařed & Sudařed PE, http://www.sudařed.com/products/pe_quickstrips.html (Aug. 17, 2007).

Well—Definition of from The American Heritage College Dictionary, 3rd Ed., p. 1531 (1993).

Bauer, K.H. et al., "Pharmazeutische Technologie", pp. 208-209 (1997).

Pinnamanemi, S. et al., "Formulation approaches for orally administered poorly soluble drugs", Pharmazie 57(5): 291-300 (2002). Chaumeil, J.C., "Micronization: A Method of Improving the

Bioavailability of Poorly Soluble Drugs", Methods and Findings in Experimental and Clinical Pharmacology 20(3): 211-215 (1998).

Page 8

(56) References Cited

OTHER PUBLICATIONS

Voigt, R. et al., "Pharmaseutische Technology für Studium und Berf", pp. 179-180 (1995).

Nanda, A. et al., "An update on taste masking technologies for oral pharmaceuticals", Indian J Pharma Sci 64(1): 10-17 (2002).

Bornschein, M. et al., "Micro- und Nanopartikeln als Arzneliestofftragersysteme unter besonderer Berucksichtigung der Herstellungsmethoden", Die Pharmazie 44(9): 585-593 (1989).

Cohen E. et al., "Modern Coating and Drying Technology", pp. 268-277 (1992).

Pharmazeutische Technologie (4th Ed.), (Bauer, K.H. et al. (eds.)), pp. 94-94, 286-287 (Georg Thieme Verlag Stuttgart1993).

Brittian, H.G., "What Is the 'Correct' Method to Use for Particle-Size Determination?," Pharmaceutical Technology 96-98 (Jul. 2001).

"More Solutions to Sticky Problems: A Guide to Getting More From Your Brookfield Viscometer," Brookfield Engineering Laboratories, Inc. (1985).

DeGrande, G., et al., "Specialized Oral Mucosal Drug Delivery Systems: Patches," Drugs and the Pharmaceutical Sciences (Swarbrick, J. (ed.)), Ch. 12, pp. 285-317 (1995).

Polymer Science and Technology (Obewele, R.O. (ed.)), pp. 1-23 (2000).

Etzler, F.M. and Sanderson, "Partilce Size Analysis: a Comparative Study of Various Methods," Part. Part. Syst. Charact. 12: 127-224 (1995).

Roddy, R.E., "A Controlled Trial of Nonoxynol 9 Film to Reduce Male-to-Female Transmission of Sexually Transmitted Diseases," New England J. Med. 339(8):504-510 (1998).

Remington's Pharmaceutical Sciences (18th Ed.) (Gennaro, A.R. (ed.)), Ch. 19, pp. 296-298 (1990).

Etzler, F.M., "Particle Size Analysis: a Comparison of Methods," Polymeric Materials: Science & Engineering 87:335-336 (2002). Patel, V.F. et al., "Advances in oral transmucosal drug delivery," J. Controlled Release 153:106-116 (2011).

"Adsorption," Kirk-Othmer Encyclopedia of Chemical Technology (4th Ed.) pp. 493-494 (Wiley 1991).

"Matrix," Webster's Third New International Dictionary of the English Language Unabridged (Gove, P.B. (ed.)) (G. & C. Merriam Company 1968).

Plastic Films (Osborn, K.R. and Jankins, W.A. (eds.), p. 89 (1992). Martinez, M.N. and Amidon, G.L., "A Mechanistic Approach to Understanding the Factors Affecting Drug Absorption: A Review of Fundamentals," J. Clin. Pharmacol. 42:620-643 (2002).

Amidon, G.L. et al., "A Theoretical Basis for a Biopharmaceutical Drug Classification: The Correlation of in Vitro Drug Product Dissolution and in Vivo Bioavailability," Pharm. Res. 12(3):413-420 (1995).

Anders, R. and Merkle, H.P., "Evaluation of laminated mucoadhesive patches for buccal drug delivery," Int. J. Pharmaceutics 49: 231-240 (1989).

Pharmaceutical Dosage Forms and Drug Delivery Systems (7th Ed) (Ansel, H.C. et al. (eds.)), p. 66 (1999).

Apicella, A. et al., "poly(ethylene oxide) (PEO) and different molecular weight PEO blends monolithic devices for drug release," Biomaterials 14(2):83-90 (1993).

Pharmazeutische Technologie (5th Ed.) (Bauer, K.H. et al. (eds.)), pp. 208-209 (Stuttgart Jena Lubeck Ulm 1997).

Bowser T.J. and Wilhelm, L.R., "Modeling Simultaneous Shrinkage and Heat and Mass Transfer of a Thin. Nonporous Film During Drying," J. Food Sci. 60(4):753-757 (1995).

Theory of pharmaceutical systems: vol. II (Carstensen, J.T. (ed.)), pp. 4-9 (1973).

Cassidy, J. P. et al., "Controlled buccal delivery of buprenorphine," J. Controlled Release 25:21-29 (1993).

Eudragit E 100, Eudragit E PO, and Eudragit E 12,5, Technical Information, Evonik Inductries AG, (2012).

Eudragit L 100 and Eudragit S 100, Technical Information, Evonik Inductries AG, (2012).

Europaisches Arzneibuch (3rd Ed.), pp. 142-143 (Deutscher Apotheker Verlag 1997).

European Pharmacopeia (3rd Ed.), p. 134 (1997).

Frankman, O. et al., "clinical Evaluation of C-Film, a Vaginal Contraceptive," J. Int. Med. Res. 3:292-296 (1975).

Friend, D.R., "Polyacrylate resin microcapsules for taste masking of antibiotics," J. Microencapsulation 9(4):469-480 (1992).

Fuller, C.S. et al., "Interactions in poly(ethylene oxide)-hydroxylpropyl methylcellulose blends," Polymer 42:9583-9592 (2001)

Save, T. et al., "Comparative Study of Buccoadhesive Formulations and Sublingual Capsules of Nifedipine," J. Pharm. Pharmacol. 46:192-195 (1994).

Save, T. and Vankitachalam, P., "Studies on Solid Dispersions of Nifedipine," Drug Development and Industrial Pharmacy 18(15):1663-1679 (1992).

Roy, G.M., "Taste Macking in Oral Pharmaceuticals," Pharmaceutical Technology, pp. 84-99 (Apr. 1994).

Guo, J.H. and Cookock, K.M., "Bioadhesive Polymer Buccal Patches for Buprenorphine Controlled Delivery: Solubility Consideration," Drug Development and Industrial Pharmacy 21(7): 2013-2019 (1995).

Mixing in the Process Industries (2nd Ed.) Harnby, N. et al. (eds.)), pp. 3, 115 (Butterworth Heinemann 1997).

Himics, R, and Pineiro, R., "The Importance of Particle Size in Liquid Coatings," Products Finishing 63(2):00329940 (1998).

Handbook of Pharmaceutical Excipients (Rowe, R. et al. (eds.)), pp. 326, 513, 522 (2009).

Ilango, R. et al., "In-Vitro Studies on Buccal strips of Glibenclamide using Chitosan," Indian J. Pharm. Sci. 59 (5):232-235 (1997).

Ishikawa, T. et al., "Preparation and Evaluation of Tablets Rapidly Disintegrating in Saliva Containing Bitter-Taste-Masked Granules by the Compression Method," Chem. Pharm. Bull. 47(10):1451-1454 (1999).

Kaya, S. and Kaya, A., "Microwave drying effects on properties of whey protein isolate edible films," J. Food Engineering, 43: 91-96 (2000).

Index and Transaction History for Ex Parte Reexamination Control No. 90/012,098, current as of.

Index of Documents for Inter Partes Review Case No. IPR2013-00316, current as of Aug. 29, 2017.

Index and Transaction History for Ex Parte Reexamination Control No. 90/012,097, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2013-00315, current as of Aug. 29, 2017.

Index and Transaction History for Inter Partes Reexamination Control No. 95/002,171, current as of Aug. 29, 2017.

Index and Transaction History for Inter Partes Reexamination Control No. 95/001,753, current as of Aug. 29, 2017.

Index and Transaction History for Inter Partes Reexamination Control No. 95/002,170, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2014-00794, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2015-00165, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2015-00167, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2015-00168, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2015-00169, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2016-00281, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2016-00282, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2016-

01111, current as of Aug. 29, 2017. Index of Documents for Inter Partes Review Case No. IPR2016-

01112, current as of Aug. 29, 2017. Index of Documents for Inter Partes Review Case No. IPR2017-

Index of Documents for Inter Partes Review Case No. IPR2017-00200, current as of Aug. 29, 2017.

Index of Documents for Inter Partes Review Case No. IPR2017-01557, current as of Aug. 29, 2017.

Page 9

(56) References Cited

OTHER PUBLICATIONS

Index of Documents for Inter Partes Review Case No. IPR2017-01582, current as of Aug. 29, 2017.

Verdampfung, Kristallisation, Trocknung (Gnielinski, V. et al., (Eds.)), pp. 161-181 (Vieweg & Sohn Verlagsgsellschaft mbH 1993). (partial English translation included.).

Giunchedi, P. and Conte, U., "Spray-drying as a preparation method of microparticulate drug delivery systems: an overview," S.T.P. Pharma. Sciences 6(4):276-290 (1995).

Guo, J.H. and Zerbe, H., "Water Soluble Film for Oral Administration," The 24th International Symposium on Controlled Release of Bioactive Materials, pp. 227-229 (Paper No. 5001-5003) (1997). The Theory and Practice of Industrial Pharmacy (3rd Ed.) (Lachman, L et al. (eds.)), pp. 47-48, 51, 57, 64, 123-127, 346-369, 453-454, 461, 470, 479, 484, 491-492, 654-655 (1986).

Physical Pharmacy (4th Ed.) (Martin, A. et al. (eds.)), pp. 423, 430-434, 453, 461, 484, 557-558, 560, 565-567 (1993).

Bioadhesive Drug Delivery Systems (Lenaerts, V. and Gurny, R. (eds.)), Ch. 6, pp. 106-136 (1990).

Introductory Polymer Chemistry (Misra, G.S. (ed.)), Ch. 6, pp. 98-118 (1993).

Nishaoka, Y. et al., "Laser Diffraction Estimation of Particle Size Distribution of Slightly Water-Soluble Drugs Coexisting with Additives: Application to Solid Dosage Forms," Chem. Pharm. Bull. 40(6):1563-1568 (1992).

Perumal, V.A. and Govender, T., "Investigating a New Approach to Film Casting for Enhanced Drug Content Uniformity in Polymeric Films," Drug Development and Industrial Pharmacy, 34:1034-1047 (2008).

Remington's Pharmaceutical Sciences (17th Ed.) (Gennaro, A.R. (ed.)), Ch. 37, pp. 713-740 (1985).

Shu, X.Z., et al., "Novel pH-sensitive citrate cross-linked chitosan film drug controlled release," Int. J. Pharmaceutics 212:19-28 (2001).

Al-Ghananeem et al., "Effect of pH on Sublingual Absorption of Oxycodone Hydrochloride", AAPS PharmSciTech; Article 23, 7(1) (2006) (http://www.aapspharmscitec.org).

Bhumkar et al., "Chitosan Reduced Gold Nanoparticles as Novel Carriers for Transmucosal Delivery of Insulin", Pharmaceutical Research; 24(8): 1415-1426 (2007).

Bowen P., "Particle Size Distribution Measurement from Millimeters to Nanometers and from Rods to Platelets", Journal of Dispersion Science and Technology; 23(5): 631-662 (2002).

Trademark Reg. No. 2,944,841—registered Apr. 26, 2005 to Reynolds Metal Co for "EZ SLIDE".

Hariharan et al., "Thin Film Technology, Orally Dissolving Film Strips (ODFS): The Final Evolution of Orally Dissolving Dosage Forms," Drug Delivery Technology; 9(2): 24-29 (2009).

Joshi et al., "Gold Nanoparticles as Carrier for Efficient Transmucosal Insulin Delivery", Langmuir, 22: 300-305 (2006). Ojeda et al., "Preparation of multifunctional glyconanoparticles as a platform for potential carbohydrate-based anticancer vaccines", Carbohydrate Research; 342: 448-459 (2007).

U.S. Department of Health and Human Services Food and Drug Administration Center for Drug Evaluation and Research (CDER) "Guidance for Industry—Incorporation of Physical-Chemical Identifiers into Solid Oral Dosage Form Drug Products for Anticounterfeiting" Silver Spring, MD; 1-8 (Jul. 9, 2009). Boo, Woong Jae, "Characterization of Thin Film Properties of

Boo, Woong Jae, "Characterization of Thin Film Properties of Melamine Based Dendrimer Nanoparticles", Thesis for Texas A&M University, Dec. 2003.

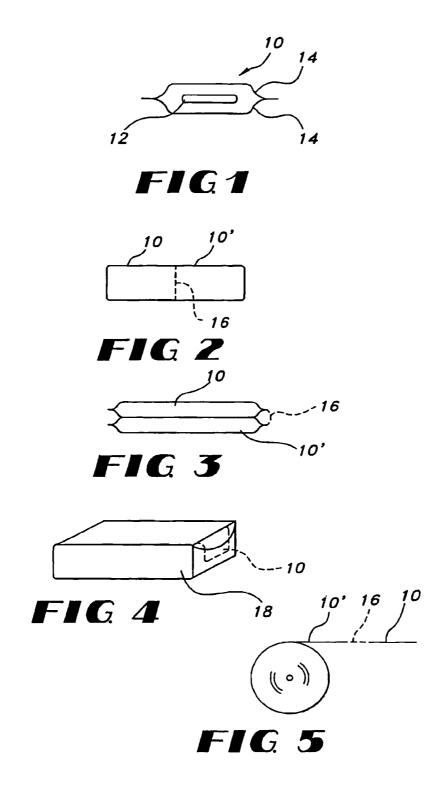
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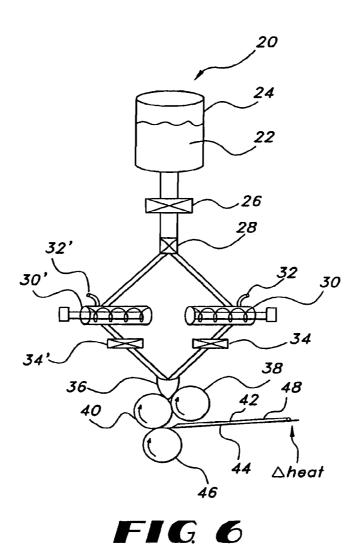


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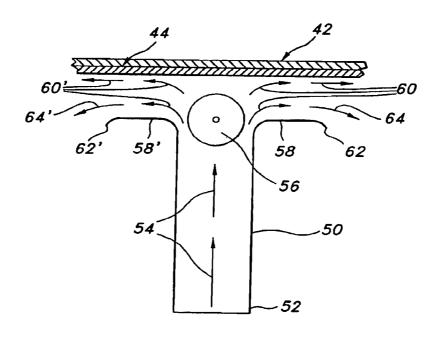


FIG 7

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UNIFORM FILMS FOR RAPID-DISSOLVE DOSAGE FORM INCORPORATING ANTI-TACKING COMPOSITIONS

CROSS-REFERENCE TO RELATED APPLICATIONS

The present application is a continuation of U.S. application Ser. No. 15/398,398, filed Jan. 4, 2017, which is a continuation of U.S. application Ser. No. 15/144,191, filed May 2, 2016, which is a continuation of U.S. application Ser. No. 14/844,810, filed Sep. 3, 2015, which is a continuation of U.S. application Ser. No. 14/284,019, filed May 21, 2014, which is a continuation of U.S. application Ser. No. 11/517, 982, filed Sep. 8, 2006, now U.S. Pat. No. 8,765,167, which claims the benefit of U.S. Provisional Application No. 60/715,528, filed Sep. 9, 2005, and is a continuation-in-part of U.S. application Ser. No. 10/074,272, filed Feb. 14, 2002, now U.S. Pat. No. 7,425,292, which claims the benefit of U.S. Provisional Application No. 60/328,868, filed Oct. 12, 2001, the contents of all of which are incorporated herein by reference.

FIELD OF THE INVENTION

The present invention relates to rapidly dissolving films incorporating anti-tacking agents and methods of their preparation. The films also may contain an active ingredient that is evenly distributed throughout the film.

BACKGROUND OF THE RELATED TECHNOLOGY

Active ingredients, such as drugs or pharmaceuticals, may be prepared in a tablet form to allow for accurate and 35 consistent dosing. However, this form of preparing and dispensing medications has many disadvantages including that a large proportion of adjuvants that must be added to obtain a size able to be handled, that a larger medication form requires additional storage space, and that dispensing 40 includes counting the tablets which has a tendency for inaccuracy. In addition, many persons, estimated to be as much as 28% of the population, have difficulty swallowing tablets. While tablets may be broken into smaller pieces or even crushed as a means of overcoming swallowing diffi- 45 culties, this is not a suitable solution for many tablet or pill forms. For example, crushing or destroying the tablet or pill form to facilitate ingestion, alone or in admixture with food, may also destroy the controlled release properties.

As an alternative to tablets and pills, films may be used to 50 carry active ingredients such as drugs, pharmaceuticals, and the like. However, historically films and the process of making drug delivery systems therefrom have suffered from a number of unfavorable characteristics that have not allowed them to be used in practice.

Films that incorporate a pharmaceutically active ingredient are disclosed in expired U.S. Pat. No. 4,136,145 to Fuchs, et al. ("Fuchs"). These films may be formed into a sheet, dried and then cut into individual doses. The Fuchs disclosure alleges the fabrication of a uniform film, which 60 includes the combination of water-soluble polymers, surfactants, flavors, sweeteners, plasticizers and drugs. These allegedly flexible films are disclosed as being useful for oral, topical or enteral use. Examples of specific uses disclosed by Fuchs include application of the films to mucosal membrane 65 areas of the body, including the mouth, rectal, vaginal, nasal and ear areas.

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Examination of films made in accordance with the process disclosed in Fuchs, however, reveals that such films suffer from the aggregation or conglomeration of particles, i.e., self-aggregation, making them inherently non-uniform. This result can be attributed to Fuchs' process parameters, which although not disclosed likely include the use of relatively long drying times, thereby facilitating intermolecular attractive forces, convection forces, air flow and the like to form such agglomeration.

The formation of agglomerates randomly distributes the film components and any active present as well. When large dosages are involved, a small change in the dimensions of the film would lead to a large difference in the amount of active per film. If such films were to include low dosages of active, it is possible that portions of the film may be substantially devoid of any active. Since sheets of film are usually cut into unit doses, certain doses may therefore be devoid of or contain an insufficient amount of active for the recommended treatment Failure to achieve a high degree of accuracy with respect to the amount of active ingredient in the cut film can be harmful to the patient. For this reason, dosage forms formed by processes such as Fuchs, would not likely meet the stringent standards of governmental or regulatory agencies, such as the U.S. Federal Drug Administration ("FDA"), relating to the variation of active in dosage forms. Currently, as required by various world regulatory authorities, dosage forms may not vary more than 10% in the amount of active present. When applied to dosage units based on films, this virtually mandates that 30 uniformity in the film be present.

The problems of self-aggregation leading to non-uniformity of a film were addressed in U.S. Pat. No. 4,849,246 to Schmidt ("Schmidt"). Schmidt specifically pointed out that the methods disclosed by Fuchs did not provide a uniform film and recognized that that the creation of a non-uniform film necessarily prevents accurate dosing, which as discussed above is especially important in the pharmaceutical area. Schmidt abandoned the idea that a mono-layer film, such as described by Fuchs, may provide an accurate dosage form and instead attempted to solve this problem by forming a multi-layered film. Moreover, his process is a multi-step process that adds expense and complexity and is not practical for commercial use.

Other U.S. patents directly addressed the problems of particle self-aggregation and non-uniformity inherent in conventional film forming techniques. In one attempt to overcome non-uniformity, U.S. Pat. No. 5,629,003 to Horstmann et al. and U.S. Pat. No. 5,948,430 to Zerbe et al, incorporated additional ingredients, i.e. gel formers and polyhydric alcohols respectively, to increase the viscosity of the film prior to drying in an effort to reduce aggregation of the components in the film. These methods have the disadvantage of requiring additional components, which translates to additional cost and manufacturing steps. Further-55 more, both methods employ the use the conventional timeconsuming drying methods such as a high-temperature airbath using a drying oven, drying tunnel, vacuum drier, or other such drying equipment. The long length of drying time aids in promoting the aggregation of the active and other adjuvant, notwithstanding the use of viscosity modifiers. Such processes also run the risk of exposing the active, i.e., a drug, or vitamin C, or other components to prolonged exposure to moisture and elevated temperatures, which may render it ineffective or even harmful.

In addition to the concerns associated with degradation of an active during extended exposure to moisture, the conventional drying methods themselves are unable to provide

uniform films. The length of heat exposure during conventional processing, often referred to as the "heat history", and the manner in which such heat is applied, have a direct effect on the formation and morphology of the resultant film product. Uniformity is particularly difficult to achieve via 5 conventional drying methods where a relatively thicker film, which is well-suited for the incorporation of a drug active, is desired. Thicker uniform films are more difficult to achieve because the surfaces of the film and the inner portions of the film do not experience the same external 10 conditions simultaneously during drying. Thus, observation of relatively thick films made from such conventional processing shows a non-uniform structure caused by convection and intermolecular forces and requires greater than 10% moisture to remain flexible. The amount of free moisture can 15 often interfere over time with the drug leading to potency issues and therefore inconsistency in the final product.

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Conventional drying methods generally include the use of forced hot air using a drying oven, drying tunnel, and the like. The difficulty in achieving a uniform film is directly 20 related to the theological properties and the process of water evaporation in the film-forming composition. When the surface of an aqueous polymer solution is contacted with a high temperature air current, such as a film-forming comimmediately evaporated forming a polymer film or skin on the surface. This seals the remainder of the aqueous filmforming composition beneath the surface, forming a barrier through which the remaining water must force itself as it is evaporated in order to achieve a dried film. As the temperature outside the film continues to increase, water vapor pressure builds up under the surface of the film, stretching the surface of the film, and ultimately ripping the film surface open allowing the water vapor to escape. As soon as the water vapor has escaped, the polymer film surface 35 reforms, and this process is repeated, until the film is completely dried. The result of the repeated destruction and reformation of the film surface is observed as a "ripple effect" which produces an uneven, and therefore non-uniform film. Frequently, depending on the polymer, a surface 40 will seal so tightly that the remaining water is difficult to remove, leading to very long drying times, higher temperatures, and higher energy costs.

Other factors, such as mixing techniques, also play a role in the manufacture of a pharmaceutical film suitable for 45 commercialization and regulatory approval. Air can be trapped in the composition during the mixing process or later during the film making process, which can leave voids in the film product as the moisture evaporates during the drying stage. The film frequently collapse around the voids 50 resulting in an uneven film surface and therefore, nonuniformity of the final film product. Uniformity is still affected even if the voids in the film caused by air bubbles do not collapse. This situation also provides a non-uniform film in that the spaces, which are not uniformly distributed, 55 are occupying area that would otherwise be occupied by the film composition. None of the above-mentioned patents either addresses or proposes a solution to the problems caused by air that has been introduced to the film.

Moreover, films go through numerous processing steps 60 prior to primary packaging, e.g., in canisters, and secondary packaging, e.g., in pouches or blister packs. The processing steps present significant challenges for the development of quality films that possess optimal film surface properties such as low coefficient of friction or high slip. Throughout 65 this process, it is important to maintain the integrity of the film from initial manufacture to final packaging. It is desir-

able, therefore, to prevent or alleviate problems that diminish the integrity of the film, such as films that soften, get tacky, adhere, dry up, or become brittle over time.

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More specifically, over-the-counter film products, such as candy and breath films, typically are packaged in canisters containing 16 film units, also referred to as strips, or higher (up to 24 or even 32 film strips per canister). The number of film strips per canister varies based on product type, active dose and packaging configuration among other considerations. When packaging multiple film strips in a canister, however, problems such as strips sticking to one another often arise.

Adherence between film strips is a common problem encountered in edible film products and may arise due to a variety of reasons. For instance, in some cases, adherence between film strips may be caused by the components used in film manufacture. Components such as flavors, plasticizers, and actives in the film can sometimes soften the film and have a detrimental effect on film quality. For example, in films having high acidulent content, the acids may exert an excessive plasticizing effect on the film. Such effect may be intensified by the hygroscopicity of some acids or other components in the film.

In some cases, adherence between film strips may be position passing through a hot air oven, the surface water is 25 caused by changes in film properties due to temperature and/or humidity changes. Some films may become tacky over time when exposed to non-optimal temperature and/or humidity conditions. This problem may be amplified for products that have a very narrow optimal temperature and/or humidity range for storage.

> Overall, films that exhibit tackiness or become tacky over time may present numerous problems. First, conversion of master rolls to daughter rolls, and further conversion to film strips becomes substantially more difficult when film is tacky. In addition, tacky film strips tend to adhere to one another when stacked in packaging, e.g., a canister. Accordingly, it becomes difficult for a user to remove a single film strip at a time from the film packaging, Overall, such adherence within the packaging decreases the aesthetics of the film strips as well as an individual consumer's ease of

> Therefore, there is a need for compositions that enable films to slide against one another, thereby providing ease of conversion, maximum storage stability and ease of consumer use, among other benefits. Further, there is a need for methods of preparing such films, which maintain the uniform distribution of components therein, thereby preventing undesired aggregations and promoting uniformity in the final film product.

SUMMARY OF THE INVENTION

In one aspect of the present invention, there is provided an edible film for delivery of an active including: an edible, water-soluble polymer; at least one anti-tacking agent selected from the group consisting of lubricants, antiadherants, glidants and combinations thereof; and an active component selected from the group consisting of cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof, wherein the film is self-support-

In another aspect of the present invention, there is provided an edible film for delivery of an active including: an edible, water-soluble polymer component which includes at least one polymer selected from hydroxypropyl cellulose, hydroxypropylmethyl cellulose, polyethylene oxide and combinations thereof; an active component selected from

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cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof; and an anti-tacking agent containing Vitamin E TPGS present in amounts of about 0.01% to about 20% by weight of the film.

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In another aspect of the present invention, there is provided an edible film for delivery of an active including: an edible, water-soluble polymer component which includes polyethylene oxide in combination with a polymer selected from hydroxypropyl cellulose, hydroxypropylmethyl cellulose and combinations thereof; and Vitamin E TPGS present in amounts sufficient to provide anti-tacking and therapeutic properties, wherein the film is self-supporting.

In some embodiments, there is provided an edible film for delivery of an active which includes: an edible, watersoluble polymer including polyethylene oxide and hydroxypropyl cellulose; polydextrose, wherein the polyethylene oxide, hydroxypropyl cellulose and polydextrose are present in a ratio of about 45:45:10; an active component selected from cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof; and at least one anti-tacking agent.

In another aspect, there is provided an edible film for delivery of an active including: (a) a self-supporting film having at least one surface, the film including: (i) an edible, 25 water-soluble polymer; and (ii) an active component selected from cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof; and (b) a coating on the at least one surface of the self-supporting film, the coating including at least one anti-tacking agent. 30

Some embodiments provide a multi-layer film for delivery of an active including: (a) at least one first film layer containing; (i) an edible, water-soluble polymer, and (ii) an anti-tacking agent; and (b) a second film layer including: (i) an edible, water-soluble polymer; and (ii) an active component selected from cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof. The first film layer is substantially in contact with the second film layer.

The present invention also provides a process for making 40 a self-supporting film having a substantially uniform distribution of components including the steps of: combining an edible, water-soluble polymer, a solvent, an active component selected from cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof and at 45 least one anti-tacking agent to form a matrix with a uniform distribution of the components; forming a self-supporting film from the matrix; providing a surface having top and bottom sides; feeding the film onto the top side of the surface; and drying the film by applying heat to the bottom 50 side of the surface.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 shows a side view of a package containing a unit 55 dosage film of the present invention.

FIG. 2 shows a top view of two adjacently coupled packages containing individual unit dosage forms of the present invention, separated by a tearable perforation.

FIG. 3 shows a side view of the adjacently coupled 60 packages of FIG. 2 arranged in a stacked configuration.

FIG. 4 shows a perspective view of a dispenser for dispensing the packaged unit dosage forms, dispenser containing the packaged unit dosage forms in a stacked configuration.

FIG. **5** is a schematic view of a roll of coupled unit dose packages of the present invention.

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FIG. 6 is a schematic view of an apparatus suitable for preparation of a pre-mix, addition of an active, and subsequent formation of the film.

FIG. 7 is a schematic view of an apparatus suitable for drying the films of the present invention.

DETAILED DESCRIPTION OF THE INVENTION

For the purposes of the present invention the term nonself-aggregating uniform heterogeneity refers to the ability of the films of the present invention, which are formed from one or more components in addition to a polar solvent, to provide a substantially reduced occurrence of, i.e. little or no, aggregation or conglomeration of components within the film as is normally experienced when films are formed by conventional drying methods such as a high-temperature air-bath using a drying oven, drying tunnel, vacuum drier, or other such drying equipment. The term heterogeneity, as used in the present invention, includes films that will incorporate a single component, such as a polymer, as well as combinations of components, such as a polymer and an active. Uniform heterogeneity includes the substantial absence of aggregates or conglomerates as is common in conventional mixing and heat drying methods used to form

Furthermore, the films of the present invention have a substantially uniform thickness, which is also not provided by the use of conventional drying methods used for drying water-based polymer systems. The absence of a uniform thickness detrimentally affects uniformity of component distribution throughout the area of a given film.

The film products of the present invention are produced by a combination of a properly selected polymer and a polar solvent, optionally including an active ingredient as well as other fillers known in the art. These films provide a non-self-aggregating uniform heterogeneity of the components within them by utilizing a selected casting or deposition method and a controlled drying process. Examples of controlled drying processes include, but are not limited to, the use of the apparatus disclosed in U.S. Pat. No. 4,631,837 to Magoon ("Magoon"), herein incorporated by reference, as well as hot air impingement across the bottom substrate and bottom heating plates. Another drying technique for obtaining the films of the present invention is controlled radiation drying, in the absence of uncontrolled air currents, such as infrared and radio frequency radiation (i.e. microwaves).

The objective of the drying process is to provide a method of drying the films that avoids complications, such as the noted "rippling" effect, that are associated with conventional drying methods and which initially dry the upper surface of the film, trapping moisture inside. In conventional oven drying methods, as the moisture trapped inside subsequently evaporates, the top surface is altered by being ripped open and then reformed. These complications are avoided by the present invention, and a uniform film is provided by drying the bottom surface of the film first or otherwise preventing the formation of polymer film formation (skin) on the top surface of the film prior to drying the depth of the film. This may be achieved by applying heat to the bottom surface of the film with substantially no top air flow, or alternatively by the introduction of controlled microwaves to evaporate the water or other polar solvent within the film, again with substantially no top air flow. Yet alternatively, drying may be achieved by using balanced fluid flow, such as balanced air flow, where the bottom and top air flows are controlled to provide a uniform film. In such a case, the air flow directed 7

at the top of the film should not create a condition which would cause movement of particles present in the wet film, due to forces generated by the air currents. Additionally, air currents directed at the bottom of the film should desirably be controlled such that the film does not lift up due to forces from the air. Uncontrolled air currents, either above or below the film, can create non-uniformity in the final film products. The humidity level of the area surrounding the top surface may also be appropriately adjusted to prevent premature closure or skinning of the polymer surface.

This manner of drying the films provides several advantages. Among these are the faster drying times and a more uniform surface of the film, as well as uniform distribution of components for any given area in the film. In addition, the faster drying time allows viscosity to quickly build within the film, further encouraging a uniform distribution of components and decrease in aggregation of components in the final film product. Desirably, the drying of the film will occur within about ten minutes or fewer, or more desirably within about five minutes or fewer.

The present invention yields exceptionally uniform film products when attention is paid to reducing the aggregation of the compositional components. By avoiding the introduction of and eliminating excessive air in the mixing process, selecting polymers and solvents to provide a controllable viscosity and by drying the film in a rapid manner from the bottom up, such films result.

The products and processes of the present invention rely on the interaction among various steps of the production of the films in order to provide films that substantially reduce the self-aggregation of the components within the films. Specifically, these steps include the particular method used to form the film, making the composition mixture to prevent air bubble inclusions, controlling the viscosity of the film forming composition and the method of drying the film. More particularly, a greater viscosity of components in the mixture is particularly useful when the active is not soluble in the selected polar solvent in order to prevent the active from settling out. However, the viscosity must not be too great as to hinder or prevent the chosen method of casting, which desirably includes reverse roll coating due to its ability to provide a film of substantially consistent thickness.

In addition to the viscosity of the film or film-forming components or matrix, there are other considerations taken into account by the present invention for achieving desirable film uniformity. For example, stable suspensions are achieved which prevent solid (such as drug particles) sedimentation in non-colloidal applications. One approach provided by the present invention is to balance the density of the particulate (ρ_p) and the liquid phase (ρ_l) and increase the viscosity of the liquid phase (μ) . For an isolated particle, Stokes law relates the terminal settling velocity (Vo) of a rigid spherical body of radius (r) in a viscous fluid, as follows:

$$V_o = (2gr^r)(\rho_p - \rho_l)/9\mu$$
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At high particle concentrations, however, the local particle concentration will affect the local viscosity and density. The viscosity of the suspension is a strong function of solids volume fraction, and particle-particle and particle-liquid interactions will further hinder settling velocity.

Stokian analyses has shown that the incorporation of a 60 third phase, dispersed air or nitrogen, for example, promotes suspension stability. Further, increasing the number of particles leads to a hindered settling effect based on the solids volume fraction. In dilute particle suspensions, the rate of sedimentation, v, can be expressed as: 65

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where κ =a constant, and ϕ is the volume fraction of the dispersed phase. More particles suspended in the liquid phase results in decreased velocity. Particle geometry is also an important factor since the particle dimensions will affect particle-particle flow interactions.

Similarly, the viscosity of the suspension is dependent on the volume fraction of dispersed solids. For dilute suspensions of non-interaction spherical particles, an expression for the suspension viscosity can be expressed as:

$$\mu/\mu_o = 1 + 2.5 \varphi$$

where μ_o is the viscosity of the continuous phase and φ is the solids volume fraction. At higher volume fractions, the viscosity of the dispersion can be expressed as

$$\mu/\mu_o = 1 + 2.5 \varphi + C_1 \varphi^2 + C_2 \varphi^3 + \dots$$

where C is a constant.

The viscosity of the liquid phase is critical and is desirably modified by customizing the liquid composition to a vis20 coelastic non-Newtonian fluid with low yield stress values. This is the equivalent of producing a high viscosity continuous phase at rest. Formation of a viscoelastic or a highly structured fluid phase provides additional resistive forces to particle sedimentation. Further, flocculation or aggregation can be controlled minimizing particle-particle interactions. The net effect would be the preservation of a homogeneous dispersed phase.

The addition of hydrocolloids to the aqueous phase of the suspension increases viscosity, may produce viscoelasticity and can impart stability depending on the type of hydrocolloid, its concentration and the particle composition, geometry, size, and volume fraction. The particle size distribution of the dispersed phase needs to be controlled by selecting the smallest realistic particle size in the high viscosity medium, i.e., <500 µm. The presence of a slight yield stress or elastic body at low shear rates may also induce permanent stability regardless of the apparent viscosity. The critical particle diameter can be calculated from the yield stress values. In the case of isolated spherical particles, the maximum shear stress developed in settling through a medium of given viscosity can be given as

$$\tau_{max}=3V\mu/2r$$
:

For pseudoplastic fluids, the viscosity in this shear stress regime may well be the zero shear rate viscosity at the Newtonian plateau.

A stable suspension is an important characteristic for the manufacture of a pre-mix composition which is to be fed into the film casting machinery film, as well as the maintenance of this stability in the wet film stage until sufficient drying has occurred to lock-in the particles and matrix into a sufficiently solid form such that uniformity is maintained. For viscoelastic fluid systems, a theology that yields stable suspensions for extended time period, such as 24 hours, must be balanced with the requirements of high-speed film casting operations. A desirable property for the films is shear thinning or pseudoplasticity, whereby the viscosity decreases with increasing shear rate. Time dependent shear effects such as thixotropy are also advantageous. Structural recovery and shear thinning behavior are important properties, as is the ability for the film to self-level as it is formed.

The rheology requirements for the inventive compositions and films are quite severe. This is due to the need to produce a stable suspension of particles, for example 30-60 wt %, in a viscoelastic fluid matrix with acceptable viscosity values throughout a broad shear rate range. During mixing, pumping, and film casting, shear rates in the range of 10-10⁵ sec. -1 may be experienced and pseudoplasticity is the preferred embodiment.

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In film casting or coating, rheology is also a defining factor with respect to the ability to form films with the desired uniformity. Shear viscosity, extensional viscosity, viscoelasticity, structural recovery will influence the quality of the film. As an illustrative example, the leveling of ⁵ shear-thinning pseudoplastic fluids has been derived as

$$\alpha^{(n-1/n)} = \alpha_o^{(n-1/n)} - ((n-1)/(2n-1))(\tau/K)^{1/n} (2\pi/\lambda)^{(3+n)/n} + \alpha_o^{(2n+1)/n} t$$

where α is the surface wave amplitude, a, is the initial amplitude, λ is the wavelength of the surface roughness, and both "n" and "K" are viscosity power law indices. In this example, leveling behavior is related to viscosity, increasing as n decreases, and decreasing with increasing K.

Desirably, the films or film-forming compositions of the present invention have a very rapid structural recovery, i.e. as the film is formed during processing, it doesn't fall apart or become discontinuous in its structure and compositional uniformity. Such very rapid structural recovery retards particle settling and sedimentation. Moreover, the films or film-forming compositions of the present invention are desirably shear-thinning pseudoplastic fluids. Such fluids with consideration of properties, such as viscosity and elasticity, promote thin film formation and uniformity.

Thus, uniformity in the mixture of components depends upon numerous variables. As described herein, viscosity of the components, the mixing techniques and the theological properties of the resultant mixed composition and wet casted film are important aspects of the present invention. Additionally, control of particle size and particle shape are further considerations. Desirably, the size of the particulate a particle size of 150 microns or less, for example 100 microns or less. Moreover, such particles may be spherical, substantially spherical, or non-spherical, such as irregularly shaped particles or ellipsoidally shaped particles. Ellipsoidally shaped particles or ellipsoids are desirable because of their ability to maintain uniformity in the film forming matrix as they tend to settle to a lesser degree as compared to spherical particles.

A number of techniques may be employed in the mixing stage to prevent bubble inclusions in the final film. To provide a composition mixture with substantially no air bubble formation in the final product, anti-foaming or surface-tension reducing agents are employed. Additionally, the 45 speed of the mixture is desirably controlled to prevent cavitation of the mixture in a manner which pulls air into the mix. Finally, air bubble reduction can further be achieved by allowing the mix to stand for a sufficient time for bubbles to escape prior to drying the film. Desirably, the inventive 50 process first forms a masterbatch of film-forming components without active ingredients such as drug particles or volatile materials such as flavor oils. The actives are added to smaller mixes of the masterbatch just prior to casting. Thus, the masterbatch pre-mix can be allowed to stand for 55 a longer time without concern for instability in drug or other

When the matrix is formed including the film-forming polymer and polar solvent in addition to any additives and the active ingredient, this may be done in a number of steps. 60 For example, the ingredients may all be added together or a pre-mix may be prepared. The advantage of a pre-mix is that all ingredients except for the active may be combined in advance, with the active added just prior to formation of the film. This is especially important for actives that may 65 degrade with prolonged exposure to water, air or another polar solvent.

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FIG. 6 shows an apparatus 20 suitable for the preparation of a pre-mix, addition of an active and subsequent formation of a film. The pre-mix or master batch 22, which includes the film-forming polymer, polar solvent, and any other additives except a drug active is added to the master batch feed tank 24. The components for pre-mix or master batch 22 are desirably formed in a mixer (not shown) prior to their addition into the master batch feed tank 24. Then a predetermined amount of the master batch is controllably fed via a first metering pump 26 and control valve 28 to either or both of the first and second mixers, 30, 30'. The present invention, however, is not limited to the use of two mixers, 30, 30', and any number of mixers may suitably be used. Moreover, the present invention is not limited to any particular sequencing of the mixers 30, 30', such as parallel sequencing as depicted in FIG. 6, and other sequencing or arrangements of mixers, such as series or combination of parallel and series, may suitably be used. The required amount of the drug or other ingredient, such as a flavor, is added to the desired mixer through an opening, 32, 32', in each of the mixers, 30, 30'. Desirably, the residence time of the pre-mix or master batch 22 is minimized in the mixers 30, 30'. While complete dispersion of the drug into the pre-mix or master batch 22 is desirable, excessive residence 25 times may result in leaching or dissolving of the drug, especially in the case for a soluble drug. Thus, the mixers 30, 30' are often smaller, i.e. lower residence times, as compared to the primary mixers (not shown) used in forming the pre-mix or master batch 22. After the drug has been blended with the master batch pre-mix for a sufficient time to provide a uniform matrix, a specific amount of the uniform matrix is then fed to the pan 36 through the second metering pumps, 34, 34'. The metering roller 38 determines the thickness of the film 42 and applies it to the application roller. The film 42 is finally formed on the substrate 44 and carried away via the support roller 46.

While the proper viscosity uniformity in mixture and stable suspension of particles, and casting method are important in the initial steps of forming the composition and film to promote uniformity, the method of drying the wet film is also important. Although these parameters and properties assist uniformity initially, a controlled rapid drying process ensures that the uniformity will be maintained until the film is dry.

The wet film is then dried using controlled bottom drying or controlled microwave drying, desirably in the absence of external air currents or heat on the top (exposed) surface of the film 48 as described herein. Controlled bottom drying or controlled microwave drying advantageously allows for vapor release from the film without the disadvantages of the prior art. Conventional convection air drying from the top is not employed because it initiates drying at the top uppermost portion of the film, thereby forming a barrier against fluid flow, such as the evaporative vapors, and thermal flow, such as the thermal energy for drying. Such dried upper portions serve as a barrier to further vapor release as the portions beneath are dried, which results in non-uniform films. As previously mentioned some top air flow can be used to aid the drying of the films of the present invention, but it must not create a condition that would cause particle movement or a rippling effect in the film, both of which would result in non-uniformity. If top air is employed, it is balanced with the bottom air drying to avoid non-uniformity and prevent film lift-up on the carrier belt. A balance top and bottom air flow may be suitable where the bottom air flow functions as the major SOURSce of drying and the top air flow is the minor SOURSce of drying. The advantage of some top air flow is

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to move the exiting vapors away from the film thereby aiding in the overall drying process. The use of any top air flow or top drying, however, must be balanced by a number of factors including, but not limited, to rheological properties of the composition and mechanical aspects of the 5 processing. Any top fluid flow, such as air, also must not overcome the inherent viscosity of the film-forming composition. In other words, the top air flow cannot break, distort or otherwise physically disturb the surface of the composition. Moreover, air velocities are desirably below the yield values of the film, i.e., below any force level that can move the liquids in the film-forming compositions. For thin or low viscosity compositions, low air velocity must be used. For thick or high viscosity compositions, higher air velocities may be used. Furthermore, air velocities are 15 desirable low so as to avoid any lifting or other movement of the film formed from the compositions.

Moreover, the films of the present invention may contain particles that are sensitive to temperature, such as flavors, which may be volatile, or drugs, which may have a low 20 degradation temperature. In such cases, the drying temperature may be decreased while increasing the drying time to adequately dry the uniform films of the present invention. Furthermore, bottom drying also tends to result in a lower internal film temperature as compared to top drying. In 25 bottom drying, the evaporating vapors more readily carry heat away from the film as compared to top drying which lowers the internal film temperature. Such lower internal film temperatures often result in decreased drug degradation and decreased loss of certain volatiles, such as flavors.

Furthermore, particles or particulates may be added to the film-forming composition or matrix after the composition or matrix is cast into a film. For example, particles may be added to the film 42 prior to the drying of the film 42. Particles may be controllably metered to the film and 35 disposed onto the film through a suitable technique, such as through the use of a doctor blade (not shown) which is a device which marginally or softly touches the surface of the film and controllably disposes the particles onto the film surface. Other suitable, but non-limiting, techniques include 40 the use of an additional roller to place the particles on the film surface, spraying the particles onto the film surface, and the like. The particles may be placed on either or both of the opposed film surfaces, i.e., the top and/or bottom film surfaces. Desirably, the particles are securably disposed onto 45 the film, such as being embedded into the film. Moreover, such particles are desirably not fully encased or fully embedded into the film, but remain exposed to the surface of the film, such as in the case where the particles are partially embedded or partially encased.

The particles may be any useful organoleptic agent, cosmetic agent, pharmaceutical agent, or combinations thereof. Desirably, the pharmaceutical agent is a tastemasked or a controlled-release pharmaceutical agent Useful cosmetic agents include breath freshening or decongestant agents, such as menthol, including menthol crystals.

Although the inventive process is not limited to any particular apparatus for the above-described desirable drying, one particular useful drying apparatus 50 is depicted in 60 FIG. 7. Drying apparatus 50 is a nozzle arrangement for directing hot fluid, such as but not limited to hot air, towards the bottom of the film 42 which is disposed on substrate 44. Hot air enters the entrance end 52 of the drying apparatus and travels vertically upward, as depicted by vectors 54, 65 towards air deflector 56. The air deflector 56 redirects the air movement to minimize upward force on the film 42. As

depicted in FIG. 7, the air is tangentially directed, as indicated by vectors 60 and 60', as the air passes by air deflector 56 and enters and travels through chamber portions 58 and 58' of the drying apparatus 50. With the hot air flow being substantially tangential to the film 42, lifting of the film as it is being dried is thereby minimized. While the air deflector 56 is depicted as a miler, other devices and geometries for deflecting air or hot fluid may suitable be used. Furthermore, the exit ends 62 and 62' of the drying apparatus 50 are flared downwardly. Such downward flaring provides a downward force or downward velocity vector, as indicated by vectors 64 and 64', which tend to provide a pulling or drag effect of the film 42 to prevent lifting of the film 42. Lifting of the film 42 may not only result in non-uniformity in the film or otherwise, but may also result in non-controlled processing of the film 42 as the film 42 and/or substrate 44 lift away from the processing equipment.

Monitoring and control of the thickness of the film also contributes to the production of a uniform film by providing a film of uniform thickness. The thickness of the film may be monitored with gauges such as Beta Gauges. A gauge may be coupled to another gauge at the end of the drying apparatus, i.e. drying oven or tunnel, to communicate through feedback loops to control and adjust the opening in the coating apparatus, resulting in control of uniform film thickness.

The film products are generally formed by combining a properly selected polymer and polar solvent, as well as any active ingredient or filler as desired. Desirably, the solvent content of the combination is at least about 30% by weight of the total combination. The matrix formed by this combination is formed into a film, desirably by roll coating, and then dried, desirably by a rapid and controlled drying process to maintain the uniformity of the film, more specifically, a non-self-aggregating uniform heterogeneity. The resulting film will desirably contain less than about 10% by weight solvent, more desirably less than about 8% by weight solvent, even more desirably less than about 6% by weight solvent and most desirably less than about 2%. The solvent may be water, a polar organic solvent including, but not limited to, ethanol, isopropanol, acetone, methylene chloride, or any combination thereof.

Consideration of the above discussed parameters, such as but not limited to theology properties, viscosity, mixing method, casting method and drying method, also impact material selection for the different components of the present invention. Furthermore, such consideration with proper material selection provides the compositions of the present invention, including a pharmaceutical and/or cosmetic dosage form or film product having no more than a 10% variance of a pharmaceutical and/or cosmetic active per unit area. In other words, the uniformity of the present invention is determined by the presence of no more than a 10% by weight of pharmaceutical and/or cosmetic variance throughorganoleptic agents include flavors and sweeteners. Useful 55 out the matrix. Desirably, the variance is less than 5% by weight, less than 2% by weight, less than 1% by weight, or less than 0.5% by weight

Film-Forming Polymers

The polymer may be water soluble, water swellable, water insoluble, or a combination of one or more either water soluble, water swellable or water insoluble polymers. The polymer may include cellulose or a cellulose derivative. Specific examples of useful water soluble polymers include, but are not limited to, polyethylene oxide (PEO), pullulan, hydroxypropylmethyl cellulose, hydroxyethyl cellulose, hydroxypropyl cellulose, polyvinyl pyrrolidone, carboxymethyl cellulose, polyvinyl alcohol, sodium aginate, polyeth-

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ylene glycol, xanthan gum, tragancanth gum, guar gum, acacia gum, arabic gum, polyacrylic acid, methylmethacrylate copolymer, carboxyvinyl copolymers, starch, gelatin, and combinations thereof. In some embodiments, combinations of PEO and a cellulosic polymer, such as hydroxypropyl cellulose, are employed. Specific examples of useful water insoluble polymers include, but are not limited to, ethyl cellulose, hydroxypropyl ethyl cellulose, cellulose acetate phthalate, hydroxypropyl methyl cellulose phthalate and combinations thereof.

As used herein the phrase "water soluble polymer" and variants thereof refer to a polymer that is at least partially soluble in water, and desirably fully or predominantly soluble in water, or absorbs water. Polymers that absorb water are often referred to as being water swellable polymers. The materials useful with the present invention may be water soluble or water swellable at room temperature and other temperatures, such as temperatures exceeding room temperature. Moreover, the materials may be water soluble 20 or water swellable at pressures less than atmospheric pressure. Desirably, the water soluble polymers are water soluble or water swellable having at least 20 percent by weight water uptake. Water swellable polymers having a 25 or greater percent by weight water uptake are also useful. Films or 25 dosage forms of the present invention formed from such water soluble polymers are desirably sufficiently water soluble to be dissolvable upon contact with bodily fluids.

Other polymers useful for incorporation into the films of the present invention include biodegradable polymers, copo- 30 lymers, block polymers and combinations thereof. Among the known useful polymers or polymer classes which meet the above criteria are: poly(glycolic acid) (PA), poly(lactic acid) (PLA), polydioxanoes, polyoxalates, poly(α -esters), polyanhydrides, polyacetates, polycaprolactones, poly(or- 35 thoesters), polyamino acids, polyaminocarbonates, polyurethanes, polycarbonates, polyamides, poly(alkyl cyanoacrylates), and mixtures and copolymers thereof. Additional useful polymers include, stereopolymers of L- and D-lactic acid, copolymers of bis(p-carboxyphenoxy) propane acid 40 and sebacic acid, sebacic acid copolymers, copolymers of caprolactone, poly(lactic acid)/poly(glycolic acid)/polyethyleneglycol copolymers, copolymers of polyurethane and (poly(lactic acid), copolymers of polyurethane and poly (lactic acid), copolymers of α -amino acids, copolymers of 45 α -amino acids and caproic acid, copolymers of α -benzyl glutamate and polyethylene glycol, copolymers of succinate and poly(glycols), polyphosphazene, polyhydroxy-alkanoates and mixtures thereof. Binary and ternary systems are contemplated.

Other specific polymers useful include those marketed under the Medisorb and Biodel trademarks. The Medisorb materials are marketed by the Dupont Company of Wilmington, Del. and are generically identified as a "lactide/ glycolide co-polymer" containing "propanoic acid, 2-hy- 55 droxy-polymer with hydroxy-polymer with hydroxyacetic acid." Four such polymers include lactide/glycolide 100L, believed to be 100% lactide having a melting point within the range of 338°-347° F. (170°-175° C.); lactide/glycolide 100L, believed to be 100% glycolide having a melting point 60 within the range of 437°-455° F. (225-235° C.); lactide/ glycolide 85/15, believed to be 85% lactide and 15% glycolide with a melting point within the range of 338°-347° F. (170°-175° C.); and lactide/glycolide 50/50, believed to be a copolymer of 50% lactide and 50% glycolide with a 65 melting point within the range of 338°. 347° F. (170°-175°

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The Biodel materials represent a family of various polyanhydrides which differ chemically.

Although a variety of different polymers may be used, it is desired to select polymers to provide a desired viscosity of the mixture prior to drying. For example, if the active or other components are not soluble in the selected solvent, a polymer that will provide a greater viscosity is desired to assist in maintaining uniformity. On the other hand, if the components are soluble in the solvent, a polymer that provides a lower viscosity may be preferred.

The polymer plays an important role in affecting the viscosity of the film. Viscosity is one property of a liquid that controls the stability of the active in an emulsion, a colloid or a suspension. Generally the viscosity of the matrix will vary from about 400 cps to about 100,000 cps, preferably from about 800 cps to about 60,000 cps, and most preferably from about 1,000 cps to about 40,000 cps. Desirably, the viscosity of the film-forming matrix will rapidly increase upon initiation of the drying process.

The viscosity may be adjusted based on the selected active depending on the other components within the matrix. For example, if the component is not soluble within the selected solvent, a proper viscosity may be selected to prevent the component from settling which would adversely affect the uniformity of the resulting film. The viscosity may be adjusted in different ways. To increase viscosity of the film matrix, the polymer may be chosen of a higher molecular weight or crosslinkers may be added, such as salts of calcium, sodium and potassium. The viscosity may also be adjusted by adjusting the temperature or by adding a viscosity increasing component. Components that will increase the viscosity or stabilize the emulsion/suspension include higher molecular weight polymers and polysaccharides and gums, which include without limitation, alginate, carageenan, hydroxypropyl methyl cellulose, locust bean gum, guar gum, xanthan gum, dextran, gum arabic, gellan gum and combinations thereof.

It has also been observed that certain polymers which when used alone would ordinarily require a plasticizer to achieve a flexible film, can be combined without a plasticizer and yet achieve flexible films. For example, HPMC and HPC when used in combination provide a flexible, strong film with the appropriate plasticity and elasticity for manufacturing and storage. No additional plasticizer or polyalcohol is needed for flexibility.

Controlled Release Films

The term "controlled release" is intended to mean the release of active at a pre-selected or desired rate. This rate will vary depending upon the application. Desirable rates include fast or immediate release profiles as well as delayed, sustained or sequential release. Combinations of release patterns, such as initial spiked release followed by lower levels of sustained release of active are contemplated. Pulsed 55 drug releases are also contemplated.

The polymers that are chosen for the films of the present invention may also be chosen to allow for controlled disintegration of the active. This may be achieved by providing a substantially water insoluble film that incorporates an active that will be released from the film over time. This may be accomplished by incorporating a variety of different soluble or insoluble polymers and may also include biodegradable polymers in combination. Alternatively, coated controlled release active particles may be incorporated into a readily soluble film matrix to achieve the controlled release property of the active inside the digestive system upon consumption.

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Films that provide a controlled release of the active are particularly useful for buccal, gingival, sublingual and vaginal applications. The films of the present invention are particularly useful where mucosal membranes or mucosal fluid is present due to their ability to readily wet and adhere to these areas.

The convenience of administering a single dose of a medication which releases active ingredients in a controlled fashion over an extended period of time as opposed to the administration of a number of single doses at regular intervals has long been recognized in the pharmaceutical arts. The advantage to the patient and clinician in having consistent and uniform blood levels of medication over an extended period of time are likewise recognized. The advantages of a variety of sustained release dosage forms are well 15 known. However, the preparation of a film that provides the controlled release of an active has advantages in addition to those well-known for controlled release tablets. For example, thin films are difficult to inadvertently aspirate and provide an increased patient compliance because they need 20 not be swallowed like a tablet. Moreover, certain embodiments of the inventive films are designed to adhere to the buccal cavity and tongue, where they controllably dissolve. Furthermore, thin films may not be crushed in the manner of controlled release tablets which is a problem leading to 25 abuse of drugs such as Oxycontin.

The actives employed in the present invention may be incorporated into the film compositions of the present invention in a controlled release form. For example, particles of drug may be coated with polymers such as ethyl cellulose or 30 polymethacrylate, commercially available under brand names such as Aquacoat ECD and Eudragit E-100, respectively. Solutions of drug may also be absorbed on such polymer materials and incorporated into the inventive film compositions. Other components such as fats and waxes, as 35 well as sweeteners and/or flavors may also be employed in such controlled release compositions.

The actives may be taste-masked prior to incorporation into the film composition, as set forth in co-pending PCT application titled, Uniform Films For Rapid Dissolve Dosage Form Incorporating Taste-Masking Compositions, (based on U.S. Provisional Application No, Express Mail Label No.: EU552991605 US of the same title, filed Sep. 27, 2003, attorney docket No. 1199-15P) the entire subject matter of which is incorporated by reference herein.

When an active is introduced to the film, the amount of active per unit area is determined by the uniform distribution of the film. For example, when the films are cut into individual dosage forms, the amount of the active in the 50 dosage form can be known with a great deal of accuracy. This is achieved because the amount of the active in a given area is substantially identical to the amount of active in an area of the same dimensions in another part of the film. The accuracy in dosage is particularly advantageous when the 55 active is a medicament, i.e. a drug.

The active components that may be incorporated into the films of the present invention include, without limitation pharmaceutical and cosmetic actives, drugs, medicaments, antigens or allergens such as ragweed pollen, spores, microorganisms including bacteria, seeds, mouthwash components such as chlorates or chlorites, flavors, fragrances, enzymes, preservatives, sweetening agents, colorants, spices, vitamins and combinations thereof.

A wide variety of medicaments, bioactive active substances and pharmaceutical compositions may be included in the dosage forms of the present invention. Examples of

useful drugs include ace-inhibitors, antianginal drugs, antiarrhythmias, anti-asthmatics, anti-cholesterolemics, analgesics, anesthetics, anti-convulsants, anti-depressants, antidiabetic agents, anti-diarrhea preparations, antidotes, antihistamines, anti-hypertensive drugs, anti-inflammatory agents, anti-lipid agents, anti-manics, anti-nauseants, antistroke agents, anti-thyroid preparations, anti-tumor drugs, anti-viral agents, acne drugs, alkaloids, amino acid preparations, anti-tussives, anti-uricemic drugs, anti-viral drugs, anabolic preparations, systemic and non-systemic anti-infective agents, anti-neoplastics, anti-parkinsonian agents, anti-rheumatic agents, appetite stimulants, biological response modifiers, blood modifiers, bone metabolism regulators, cardiovascular agents, central nervous system stimulates, cholinesterase inhibitors, contraceptives, decongestants, dietary supplements, dopamine receptor agonists, endometriosis management agents, enzymes, erectile dysfunction therapies, fertility agents, gastrointestinal agents, homeopathic remedies, hormones, hypercalcemia and hypocalcemia management agents, immunomodulators, immunosuppressives, migraine preparations, motion sickness treatments, muscle relaxants, obesity management agents, osteoporosis preparations, oxytocics, parasympatholytics, parasympathomimetics, prostaglandins, psychotherapeutic agents, respiratory agents, sedatives, smoking cessation aids such as bromocryptine and nicotine, sympatholytics, tremor preparations, urinary tract agents, vasodilators, laxatives, antacids, ion exchange resins, antipyretics, appetite suppressants, expectorants, anti-anxiety agents, anti-ulcer agents, anti-inflammatory substances, coronary dilators, cerebral dilators, peripheral vasodilators, psycho-tropics, stimulants, anti-hypertensive drugs, vasoconstrictors, migraine treatments, antibiotics, tranquilizers, anti-psychotics, anti-tumor drugs, anti-coagulants, antithrombotic drugs, hypnotics, anti-emetics, anti-nauseants, anti-convulsants, neuromuscular drugs, hyper- and hypoglycemic agents, thyroid and anti-thyroid preparations, diuretics, anti-spasmodics, terine relaxants, anti-obesity drugs, erythropoietic drugs, anti-asthmatics, cough suppressants, mucolytics, DNA and genetic modifying drugs, and combinations thereof.

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Examples of medicating active ingredients contemplated for use in the present invention include antacids, $\rm H_2$ -antagonists, and analgesics. For example, antacid dosages can be prepared using the ingredients calcium carbonate alone or in combination with magnesium hydroxide, and/or aluminum hydroxide. Moreover, antacids can be used in combination with Hz-antagonists.

Analgesics include opiates and opiate derivatives, such as oxycodone (available as Oxycontin®), ibuprofen, aspirin, acetaminophen, and combinations thereof that may optionally include caffeine.

Other preferred drugs for other preferred active ingredients for use in the present invention include anti-diarrheals such as immodium AD, anti-histamines, anti-tussives, decongestants, vitamins, and breath fresheners. Suitable vitamins contemplated for use herein include any conventionally known vitamins, such as, but not limited to, Vitamins A, B, C and E. Common drugs used alone or in combination for colds, pain, fever, cough, congestion, runny nose and allergies, such as acetaminophen, chlorpheniramine maleate, dextromethorphan, pseudoephedrine HQ and diphenhydramine may be included in the film compositions of the present invention.

Also contemplated for use herein are anxiolytics such as alprazolam (available as Xanax®); anti-psychotics such as clozopin (available as Clozaril®) and haloperidol (available

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as Haldol®); non-steroidal anti-inflammatories (NSAID's) such as dicyclofenacs (available as Voltaren®) and etodolac (available as Lodine®), anti-histamines such as loratadine (available as Claritin®), astemizole (available as HismanalTM), nabumetone (available as Relafen®), and Clemas- 5 tine (available as Tavist®); anti-emetics such as granisetron hydrochloride (available as Kytril®) and nabilone (available as CesametTM); bronchodilators such as Bentolin®, albuterol sulfate (available as Proventil®); anti-depressants such as fluoxetine hydrochloride (available as Prozac®), 10 sertraline hydrochloride (available as Zoloft®), and paroxtine hydrochloride (available as Paxil®); anti-migraines such as Imigra®, ACE-inhibitors such as enalaprilat (available as Vasotec®), captopril (available as Capoten®) and lisinopril (available as Zestril®); anti-Alzheimer's agents, 15 such as nicergoline; and Ca^H-antagonists such as nifedipine (available as Procardia® and Adalat®), and verapamil hydrochloride (available as Calan®).

Erectile dysfunction therapies include, but are not limited to, drugs for facilitating blood flow to the penis, and for 20 effecting autonomic nervous activities, such as increasing parasympathetic (cholinergic) and decreasing sympathetic (adrenersic) activities. Useful non-limiting drugs include sildenafils, such as Viagra®, tadalafils, such as Cialis®, vardenafils, apomorphines, such as Uprima®, yohimbine 25 hydrochlorides such as Aphrodyne®, and alprostadils such as Caverject®.

The popular H₂-antagonists which are contemplated for use in the present invention include cimetidine, ranitidine hydrochloride, famotidine, nizatidien, ebrotidine, mifentidine, roxatidine, pisatidine and aceroxatidine.

Active antacid ingredients include, but are not limited to, the following: aluminum hydroxide, dihydroxyaluminum aminoacetate, aminoacetic acid, aluminum phosphate, dihydroxyaluminum sodium carbonate, bicarbonate, bismuth 35 aluminate, bismuth carbonate, bismuth subcarbonate, bismuth subgallate, bismuth subnitrate, bismuth subsilysilate, calcium carbonate, calcium phosphate, citrate ion (acid or salt), amino acetic acid, hydrate magnesium aluminate sulfate, magnesium glycinate, magnesium hydroxide, magnesium oxide, magnesium trisilicate, milk solids, aluminum mono-ordibasic calcium phosphate, tricalcium phosphate, potassium bicarbonate, sodium tartrate, sodium bicarbonate, magnesium aluminosilicates, tartaric acids and salts. 45

The pharmaceutically active agents employed in the present invention may include allergens or antigens, such as, but not limited to, plant pollens from grasses, trees, or ragweed; animal danders, which are tiny scales shed from the skin and hair of cars and other furred animals; insects, such as house 50 dust mites, bees, and wasps; and drugs, such as penicillin.

Botanicals, herbals and minerals also may be added to the film. Examples of botanicals include, without limitation: roots; barks; leaves; stems; flowers; fruits; tobacco; sunflower seeds; snuff; and combinations thereof.

An anti-oxidant may also be added to the film to prevent the degradation of an active, especially where the active is photosensitive.

The bioactive active substances employed in the present invention may include beneficial bacteria. More specifically, 60 certain bacteria normally exist on the surface of the tongue and in the back of the throat. Such bacteria assist in the digestion of food by breaking down proteins found in the food. It may be desirable, therefore, to incorporate these bacteria into the oral film products of the present invention. 65

It also may be desirable to include actives for treating breath malodor and related oral care conditions, such as 18

actives which are effective in suppressing microorganisms. Because breath malodor can be caused by the presence of anaerobic bacteria in the oral cavity, which generate volatile sulfur compounds, components that suppress such microorganisms may be desirable. Examples of such components include antimicrobials such as triclosan, chlorine dioxide, chlorates, and chlorites, among others. The use of chlorites, particularly sodium chlorite, in oral care compositions such as mouthrinses and toothpastes is taught in U.S. Pat. Nos. 6,251,372, 6,132,702, 6,077,502, and U.S. Publication No. 2003/0129144, all of which are incorporated herein by reference. Such components are incorporated in amounts effective to treat malodor and related oral conditions.

Cosmetic active agents may include breath freshening compounds like menthol, other flavors or fragrances, especially those used for oral hygiene, as well as actives used in dental and oral cleansing such as quaternary ammonium bases. The effect of flavors may be enhanced using flavor enhancers like tartaric acid, citric acid, vanillin, or the like.

Also color additives can be used in preparing the films. Such color additives include food, drug and cosmetic colors (FD&C), drug and cosmetic colors (D&C), or external drug and cosmetic colors (Ext. D&C). These colors are dyes, their corresponding lakes, and certain natural and derived colorants. Lakes are dyes absorbed on aluminum hydroxide.

Other examples of coloring agents include known azo dyes, organic or inorganic pigments, or coloring agents of natural origin. Inorganic pigments are preferred, such as the oxides or iron or titanium, these oxides, being added in concentrations ranging from about 0.001 to about 10%, and preferably about 0.5 to about 3%, based on the weight of all the components.

Flavors may be chosen from natural and synthetic flavoring liquids. An illustrative list of such agents includes volatile oils, synthetic flavor oils, flavoring aromatics, oils, liquids, oleoresins or extracts derived from plants, leaves, flowers, fruits, stems and combinations thereof. A nonlimiting representative list of examples includes mint oils, cocoa, and citrus oils such as lemon, orange, grape, lime and grapefruit and fruit essences including apple, pear, peach, grape, strawberry, raspberry, cherry, plum, pineapple, apricot or other fruit flavors.

The films containing flavorings may be added to provide a hot or cold flavored drink or soup. These flavorings include, without limitation, tea and soup flavorings such as beef and chicken.

Other useful flavorings include aldehydes and esters such as benzaldehyde (cherry, almond), citral i.e., aiphacitral (lemon, lime), neral, i.e., beta-citral (lemon, lime), decanal (orange, lemon), aldehyde C-8 (citrus fruits), aldehyde C-9 (citrus fruits), aldehyde C-12 (citrus fruits), tolyl aldehyde (cherry, almond), 2,6-dimethyloctanol (green fruit), and 2-dodecenal (citrus, mandarin), combinations thereof and the like.

The sweeteners may be chosen from the following non-limiting list: glucose (corn syrup), dextrose, invert sugar, fructose, and combinations thereof; saccharin and its various salts such as the sodium salt; dipeptide sweeteners such as aspartame; dihydrochalcone compounds, glycyrrhizin; *Stevia Rebaudiana* (Stevioside); chloro derivatives of sucrose such as sucralose; sugar alcohols such as sorbitol, mannitol, xylitol, and the like. Also contemplated are hydrogenated starch hydrolysates and the synthetic sweetener 3,6-dihydro-6-methyl-1-1-1,2,3-oxathiazin-4-one-2,2-dioxide, particularly the potassium salt (acesulfame-K), and sodium and calcium sals thereof, and natural intensive sweeteners, such as Lo Han Kuo. Other sweeteners may also be used.

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When the active is combined with the polymer in the solvent, the type of matrix that is formed depends on the solubilities of the active and the polymer. If the active and/or polymer are soluble in the selected solvent, this may form a solution. However, if the components are not soluble, the matrix may be classified as an emulsion, a colloid, or a suspension.

Dosages

The film products of the present invention are capable of accommodating a wide range of amounts of the active 10 ingredient. The films are capable of providing an accurate dosage amount (determined by the size of the film and concentration of the active in the original polymer/water combination) regardless of whether the required dosage is high or extremely low. Therefore, depending on the type of 15 active or pharmaceutical composition that is incorporated into the film, the active amount may be as high as about 300 mg, desirably up to about 150 mg or as low as the microgram range, or any amount therebetween.

The film products and methods of the present invention 20 are well suited for high potency, low dosage drugs. This is accomplished through the high degree of uniformity of the films. Therefore, low dosage drugs, particularly more potent racemic mixtures of actives are desirable.

Anti-Tacking Compositions

It is useful to add anti-tacking agents, such as lubricants, antiadherants and glidants to the film compositions of the present invention. Anti-tacking agents assist in the flow characteristics of the material, for example, by reducing sticking to the die in extrusion processes and reducing 30 sticking to the roof of the mouth during administration of the dosage form.

During consumption of films, particles tend to adhere to the roof of the mouth. This is undesirable for films containing bitter drugs, such as, for example, dextromethorphan, 35 because the adhered particles elude drug, which increases the amount of bitterness detected by the user. Addition of an anti-tacking agent to the films reduces adherence to the roof of the mouth, thereby effectively reducing the bitterness that may be detected by a user during consumption.

Anti-taking agents also may impart reduced film-to-film coefficient of friction, thereby reducing the problem of film dosage units, i.e., strips, adhering to one another. More specifically, in many types of film packaging, strips are stacked against one another. The incorporation of anti- 45 tacking agents may permit the individual strips to slide smoothly against one another as each unit is removed from the packaging.

Examples of suitable lubricants for use as anti-tacking agents include, but are not limited to: stearates, such as 50 magnesium stearate, calcium stearate, and sodium stearate; stearic acid; sterotex; talc; waxes; stearowet; boric acid; sodium benzoate; sodium acetate; sodium chloride; DL-Leucine; Carbowax 4000; Carbowax 6000; sodium oleate; sodium lauryl sulfate; magnesium lauryl sulfate; and com- 55 binations thereof.

Examples of suitable antiadherants include, but are not limited to: tale; cornstarch; Cab-O-Sil; syloid; DL-Leucine; sodium lauryl sulfate; metallic stearates; and combinations thereof. Examples of suitable glidants include, but are not 60 limited to: tale; cornstarch; Cab-O-Sil; syloid; aerosol; and combinations thereof.

Some embodiments of the present invention include fats and/or waxes as anti-tacking agents.

Vitamin E is another suitable anti-tacking agent for use in 65 some embodiments of the present invention. Vitamin E may serve as both an anti-tacking agent and an active component

in the film. Desirably, Vitamin E TPGS (d-alpha tocopheryl polyethylene glycol 1000 succinate) is employed. Vitamin E TPGS is a water-soluble form of Vitamin E derived from natural sources. As compared to other forms, Vitamin E TPGS is easily absorbed. Further, Vitamin E TPGS imparts practically no taste to film. Vitamin E TPGS may be

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employed in solution, such as, for example 10% or 20% solution with water. Vitamin E TPGS is particularly useful in reducing the stickiness of the films and the tendency to adhere to the roof of the user's mouth. Vitamin E may be present in amounts of about 0.01% to about 20% by weight of the composition.

Anti-tacking agents generally are present in amounts of about 0.01% to about 20% by weight of the film composition. More specifically, anti-tacking agents may be present in amounts of about 0.01% to about 10% by weight of the film composition, and even more specifically, about 0.25% to about 5% by weight of the film composition.

Combinations of anti-tacking agents also may be employed. For instance, in some embodiments of the present invention, a combination of a stearate, such as magnesium stearate, and silica may be used. SIPERNAT 500LS, which is a silica product having a 4.5 µm mean particle size, is suitable for use herein (commercially available from Degussa). Combinations of magnesium stearate and silica may provide improved glidant properties, i.e., assist film strips in sliding smoothly against one another in packaging. Accordingly, in some embodiments, magnesium stearate may be present in amounts of about 0.1% to about 2.5% by weight of the film composition and silica may be present in amounts of about 0.1% to about 1.5% by weight of the film composition. Such combination of anti-tacking agents may be useful in a variety of films containing different flavors and/or actives.

In some embodiments, anti-tacking agents may be included in the film composition itself. For example, single or multi-layer films including anti-tacking agents may be formed. Multi-layer films, for example, may include two, three or more layers of film substantially in contact with one another. In some embodiments, the film layers may be laminated to one another. Anti-tacking agents may be present in one or more of the layers of the multi-layer film. For example, some embodiments may include a bi-layer film in which anti-tacking agents are present in one of the two film layers. Some embodiments may include a three-layer film in which anti-tacking agents are present in each of the outer layers but not in the inner, or middle, layer of the three-layer film. In accordance therewith, a variety of different combinations of layers may be formed.

Alternatively, in some embodiments, anti-tacking agents may be included in a composition that is used to coat the external surfaces of the film. For instance, anti-tacking agents may be applied to the film in the form of a wet or dry coating, such as, for example, a sugared or sugar-free coating. The film may be coated with the anti-tacking agents in any conventional manner, such as, but not limited to, dip coating, spray coating, dusting, or fluidized bed. One or more film surfaces may be coated. In some embodiments, the anti-tacking coating may be applied to a substrate, such as a backing for the film, rather than directly to the film itself. When the film is removed from the backing, the anti-tacking coating may adhere to the film. Anti-Foaming and Defoaming Compostions

Anti-foaming and/or de-foaming components may also be used with the films of the present invention. These components aid in the removal of air, such as entrapped air, from

the film-forming compositions. As described above, such

entrapped air may lead to non-uniform films. Simethicone is one particularly useful anti-foaming and/or de-foaming agent. The present invention, however, is not so limited and

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other anti-foam and/or de-foaming agents may suitable be used.

Simethicone is generally used in the medical field as a treatment for gas or colic in babies. Simethicone is a mixture of fully methylated linear siloxane polymers containing repeating units of polydimethylsiloxane which is stabilized with trimethylsiloxy end-blocking unites, and silicon dioxide. It usually contains 90.5-99% polymethylsiloxane and 4-7% silicon dioxide. The mixture is a gray, translucent, viscous fluid which is insoluble in water.

When dispersed in water, simethicone will spread across the surface, forming a thin film of low surface tension. In 15 this way, simethicone reduces the surface tension of bubbles air located in the solution, such as foam bubbles, causing their collapse. The function of simethicone mimics the dual action of oil and alcohol in water. For example, in an oily solution any trapped air bubbles will ascend to the surface 20 and dissipate more quickly and easily, because an oily liquid has a lighter density compared to a water solution. On the other hand, an alcohol/water mixture is known to lower water density as well as lower the water's surface tension. So, any air bubbles trapped inside this mixture solution will 25 also be easily dissipated. Simethicone solution provides both of these advantages. It lowers the surface energy of any air bubbles that trapped inside the aqueous solution, as well as lowering the surface tension of the aqueous solution. As the result of this unique functionality, simethicone has an excel- 30 lent anti-foaming property that can be used for physiological processes (anti-gas in stomach) as well as any for external processes that require the removal of air bubbles from a product.

In order to prevent the formation of air bubbles in the 35 films of the present invention, the mixing step can be performed under vacuum. However, as soon as the mixing step is completed, and the film solution is returned to the normal atmosphere condition, air will be re-introduced into or contacted with the mixture. In many cases, tiny air 40 bubbles will be again trapped inside this polymeric viscous solution. The incorporation of simethicone into the film-forming composition either substantially reduces or eliminates the formation of air bubbles.

Simethicone may be added to the film-forming mixture as 45 an anti-foaming agent in an amount from about 0.01 weight percent to about 5.0 weight percent, more desirably from about 0.05 weight percent to about 2.5 weight percent, and most desirably from about 0.1 weight percent to about 1.0 weight percent.

Optional Components

A variety of other components and fillers may also be added to the films of the present invention. These may include, without limitation, surfactants; plasticizers which assist in compatibilizing the components within the mixture; 55 polyalcohols; anti-foaming agents, such as silicone-containing compounds, which promote a smoother film surface by releasing oxygen from the film; and thermo-setting gels such as pectin, carageenan, and gelatin, which help in maintaining the dispersion of components.

The variety of additives that can be incorporated into the inventive compositions may provide a variety of different functions. Examples of classes of additives include excipients, lubricants, buffering agents, stabilizers, blowing agents, pigments, coloring agents, fillers, bulking agents, 65 sweetening agents, flavoring agents, fragrances, release modifiers, adjuvants, plasticizers, flow accelerators, mold

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release agents, polyols, granulating agents, diluents, binders, buffers, absorbents, glidants, adhesives, anti-adherents, acidulants, softeners, resins, demulcents, solvents, surfactants, emulsifiers, elastomers and mixtures thereof. These additives may be added with the active ingredient(s).

Useful additives include, for example, gelatin, vegetable proteins such as sunflower protein, sovbean proteins, cotton seed proteins, peanut proteins, grape seed proteins, whey proteins, whey protein isolates, blood proteins, egg proteins, acrylated proteins, water-soluble polysaccharides such as alginates, carrageenans, guar gum, agar-agar, xanthan gum, gellan gum, gum arabic and related gums (gum ghatti, gum karaya, gum tragancanth), pectin, water-soluble derivatives of cellulose: alkylcelluloses hydroxyalkylcelluloses and hydroxyalkylalkylcelluloses, such as methylcelulose, hydroxymethylcellulose, hydroxyethylcellulose, hydroxypropylcellulose, hydroxyethylmethylcellulose, hydroxypropylmethylcellulose, hydroxybutylmethylcellulose, cellulose esters and hydroxyalkylcellulose esters such as cellulose acetate phthalate (CAP), hydroxypropylmethylcellulose (HPMC); carboxyalkylcelluloses, carboxyalkylalkylcelluloses, carboxyalkylcellulose esters such as carboxymethylcellulose and their alkali metal salts; water-soluble synthetic polymers such as polyacrylic acids and polyacrylic acid esters, polymethacrylic acids and polymethacrylic acid esters, polyvinylacetates, polyvinylalcohols, polyvinylacetatephthalates (PVAP), polyvinylpyrrolidone (PVP), PVY/ vinyl acetate copolymer, and polycrotonic acids; also suitable are phthalated gelatin, gelatin succinate, crosslinked gelatin, shellac, water soluble chemical derivatives of starch, cationically modified acrylates and methacrylates possessing, for example, a tertiary or quaternary amino group, such as the diethylaminoethyl group, which may be quaternized if desired; and other similar polymers.

Such extenders may optionally be added in any desired amount desirably within the range of up to about 80%, desirably about 3% to 50% and more desirably within the range of 3% to 20% based on the weight of all components.

Further additives may be inorganic fillers, such as the oxides of magnesium aluminum, silicon, titanium, etc. desirably in a concentration range of about 0.02% to about 3% by weight and desirably about 0.02% to about 1% based on the weight of all components.

Further examples of additives are plasticizers which include polyalkylene oxides, such as polyethylene glycols, polypopylene glycols, polyethylene-propylene glycols, organic plasticizers with low molecular weights, such as glycerol, glycerol monoacetate, diacetate or triacetate, triacetin, polysorbate, cetyl alcohol, propylene glycol, sorbitol, sodium diethylsulfosuccinate, triethyl citrate, tributyl citrate, and the like, added in concentrations ranging from about 0.5% to about 30%, and desirably ranging from about 0.5% to about 20% based on the weight of the polymer.

There may further be added compounds to improve the flow properties of the starch material such as animal or vegetable fats, desirably in their hydrogenated form, especially those which are solid at room temperature. These fats desirably have a melting point of 50° C. or higher. Preferred are tri-glycerides with C₁₂-, C₁₄-, C₁₆-, C₁₈-, C₂₀-and C₂₂-fatty acids. These fats can be added alone without adding extenders or plasticizers and can be advantageously added alone or together with mono- and/or di-glycerides or phosphatides, especially lecithin. The mono- and di-glycerides are desirably derived from the types of fats described above, i.e. with C₁₂-, C₁₄-, C₁₆-, C₁₈-, C₂₀- and C₂₂-fatty acids.

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The total amounts used of the fats, mono-, di-glycerides and/or lecithins are up to about 5% and preferably within the range of about 0.5% to about 2% by weight of the total composition

It is further useful to add silicon dioxide, calcium silicate, ⁵ or titanium dioxide in a concentration of about 0.02% to about 1% by weight of the total composition. These compounds act as texturizing agents.

These additives are to be used in amounts sufficient to achieve their intended purpose. Generally, the combination of certain of these additives will alter the overall release profile of the active ingredient and can be used to modify, i.e. impede or accelerate the release.

Lecithin is one surface active agent for use in the present 15 invention. Lecithin can be included in the feedstock in an amount of from about 0.25% to about 2.00% by weight. Other surface active agents, i.e. surfactants, include, but are not limited to, cetyl alcohol, sodium lauryl sulfate, the SpansTM and TweensTM which are commercially available 20 from ICI Americas, Inc. Ethoxylated oils, including ethoxylated castor oils, such as Cremophor® EL which is commercially available from BASF, are also useful. CarbowaxTM is yet another modifier which is very useful in the present invention. TweensTM or combinations of surface 25 active agents may be used to achieve the desired hydrophilic-lipophilic balance ("HLB"). The present invention, however, does not require the use of a surfactant and films or film-forming compositions of the present invention may be essentially free of a surfactant while still providing the 30 desirable uniformity features of the present invention.

It may be further useful to add polydextrose to the films of the present invention. Polydextrose serves as a filler and solubility enhancer, i.e., it increases the dissolution time of the films in the oral cavity.

As additional modifiers which enhance the procedure and product of the present invention are identified, Applicants intend to include all such additional modifiers within the scope of the invention claimed herein.

Other ingredients include binders which contribute to the 40 ease of formation and general quality of the films. Non-limiting examples of binders include starches, pregelatinize starches, gelatin, polyvinylpyrrolidone, methylcellulose, sodium carboxymethylcellulose, ethylcellulose, polyacrylamides, polyvinyloxoazolidone, and polyvinylalcohols.

45 Forming the Film

The films of the present invention must be formed into a sheet prior to drying. After the desired components are combined to form a multi-component matrix, including the polymer, water, and an active or other components as 50 desired, the combination is formed into a sheet or film, by any method known in the art such as extrusion, coating, spreading, casting or drawing the multi-component matrix. If a multi-layered film is desired, this may be accomplished by co-extruding more than one combination of components 55 which may be of the same or different composition. A multi-layered film may also be achieved by coating, spreading, or casting a combination onto an already formed film layer.

Although a variety of different film-forming techniques 60 may be used, it is desirable to select a method that will provide a flexible film, such as reverse roll coating. The flexibility of the film allows for the sheets of film to be rolled and transported for storage or prior to being cut into individual dosage forms. Desirably, the films will also be 65 self-supporting or in other words able to maintain their integrity and structure in the absence of a separate support.

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Furthermore, the films of the present invention may be selected of materials that are edible or ingestible.

Coating or casting methods are particularly useful for the purpose of forming the films of the present invention. Specific examples include reverse roll coating, gravure coating, immersion or dip coating, metering rod or meyer bar coating, slot die or extrusion coating, gap or knife over roll coating, air knife coating, curtain coating, or combinations thereof, especially when a multi-layered film is desired.

Roll coating, or more specifically reverse roll coating, is particularly desired when forming films in accordance with the present invention. This procedure provides excellent control and uniformity of the resulting films, which is desired in the present invention. In this procedure, the coating material is measured onto the applicator roller by the precision setting of the gap between the upper metering roller and the application roller below it. The coating is transferred from the application roller to the substrate as it passes around the support roller adjacent to the application roller. Both three roll and four roll processes are common.

The gravure coating process relies on an engraved roller running in a coating bath, which fills the engraved dots or lines of the roller with the coating material. The excess coating on the roller is wiped off by a doctor blade and the coating is then deposited onto the substrate as it passes between the engraved roller and a pressure roller.

Offset Gravure is common, where the coating is deposited on an intermediate roller before transfer to the substrate.

In the simple process of immersion or dip coating, the substrate is dipped into a bath of the coating, which is normally of a low viscosity to enable the coating to run back into the bath as the substrate emerges.

In the metering rod coating process, an excess of the coating is deposited onto the substrate as it passes over the bath roller. The wire-wound metering rod, sometimes known as a Meyer Bar, allows the desired quantity of the coating to remain on the substrate. The quantity is determined by the diameter of the wire used on the rod.

In the slot die process, the coating is squeezed out by gravity or under pressure through a slot and onto the substrate. If the coating is 100% solids, the process is termed "Extrusion" and in this case, the line speed is frequently much faster than the speed of the extrusion. This enables coatings to be considerably thinner than the width of the slot.

The gap or knife over roll process relies on a coating being applied to the substrate which then passes through a "gap" between a "knife" and a support roller. As the coating and substrate pass through, the excess is scraped off.

Air knife coating is where the coating is applied to the substrate and the excess is "blown off" by a powerful jet from the air knife. This procedure is useful for aqueous coatings.

In the curtain coating process, a bath with a slot in the base allows a continuous curtain of the coating to fall into the gap between two conveyors. The object to be coated is passed along the conveyor at a controlled speed and so receives the coating on its upper face.

Drying the Film

The drying step is also a contributing factor with regard to maintaining the uniformity of the film composition. A controlled drying process is particularly important when, in the absence of a viscosity increasing composition or a composition in which the viscosity is controlled, for example by the selection of the polymer, the components within the film may have an increased tendency to aggregate or conglomerate. An alternative method of forming a film

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with an accurate dosage, that would not necessitate the controlled drying process, would be to cast the films on a predetermined well. With this method, although the components may aggregate, this will not result in the migration of the active to an adjacent dosage form, since each well may 5 define the dosage unit per se.

When a controlled or rapid drying process is desired, this may be through a variety of methods. A variety of methods may be used including those that require the application of heat. The liquid carriers are removed from the film in a 10 manner such that the uniformity, or more specifically, the non-self-aggregating uniform heterogeneity, that is obtained in the wet film is maintained.

Desirably, the film is dried from the bottom of the film to the top of the film. Desirably, substantially no air flow is 15 present across the top of the film during its initial setting period, during which a solid, visco-elastic structure is formed. This can take place within the first few minutes, e.g. about the first 0.5 to about 4.0 minutes of the drying process. Controlling the drying in this manner, prevents the destruc- 20 tion and reformation of the film's top surface, which results from conventional drying methods. This is accomplished by forming the film and placing it on the top side of a surface having top and bottom sides. Then, heat is initially applied to the bottom side of the film to provide the necessary energy to evaporate or otherwise remove the liquid carrier. The films dried in this manner dry more quickly and evenly as compared to air-dried films, or those dried by conventional drying means. In contrast to an air-dried film that dries first at the top and edges, the films dried by applying heat to the 30 bottom dry simultaneously at the center as well as at the edges. This also prevents settling of ingredients that occurs with films dried by conventional means.

The temperature at which the films are dried is about 100° C. or less, desirably about 90° C. or less, and most desirably 35 about 80° C. or less.

Another method of controlling the drying process, which may be used alone or in combination with other controlled methods as disclosed above includes controlling and modifying the humidity within the drying apparatus where the film is being dried. In this manner, the premature drying of the top surface of the film is avoided.

The film compositions in many cases serve as mucoadhesives themselves.

The films may be applied under or to the tongue of the mammal. When this is desired, a specific film shape, corresponding to the shape of the tongue may be preferred. Therefore the film may be cut to a shape where the side of

Additionally, it has also been discovered that the length of drying time can be properly controlled, i.e. balanced with the heat sensitivity and volatility of the components, and particularly the flavor oils and drugs. The amount of energy, temperature and length and speed of the conveyor can be balanced to accommodate such actives and to minimize loss, degradation or ineffectiveness in the final film.

A specific example of an appropriate drying method is 50 that disclosed by Magoon. Magoon is specifically directed toward a method of drying fruit pulp. However, the present inventors have adapted this process toward the preparation of thin films.

The method and apparatus of Magoon are based on an 55 interesting property of water. Although water transmits energy by conduction and convection both within and to its surroundings, water only radiates energy within and to water. Therefore, the apparatus of Magoon includes a surface onto which the fruit pulp is placed that is transparent to 60 infrared radiation. The underside of the surface is in contact with a temperature controlled water bath. The water bath temperature is desirably controlled at a temperature slightly below the boiling temperature of water. When the wet fruit pulp is placed on the surface of the apparatus, this creates a 65 "refractance window." This means that infrared energy is permitted to radiate through the surface only to the area on

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the surface occupied by the fruit pulp, and only until the fruit pulp is dry. The apparatus of Magoon provides the films of the present invention with an efficient drying time reducing the instance of aggregation of the components of the film.

The films may initially have a thickness of about 500 μ m to about 1,500 μ m, or about 20 mils to about 60 mils, and when dried have a thickness from about 3 μ m to about 250 μ m, or about 0.1 mils to about 10 mils. Desirably, the dried films will have a thickness of about 2 mils to about 8 mils, and more desirably, from about 3 mils to about 6 mils. Uses of Thin Films

The thin films of the present invention are well suited for many uses. The high degree of uniformity of the components of the film makes them particularly well suited for incorporating pharmaceuticals. Furthermore, the polymers used in construction of the films may be chosen to allow for a range of disintegration times for the films. A variation or extension in the time over which a film will disintegrate may achieve control over the rate that the active is released, which may allow for a sustained release delivery system. In addition, the films may be used for the administration of an active to any of several body surfaces, especially those including mucous membranes, such as oral, anal, vaginal, opthalmological, the surface of a wound, either on a skin surface or within a body such as during surgery, and similar surfaces.

The films may be used to orally administer an active. This is accomplished by preparing the films as described above and introducing them to the oral cavity of a mammal. This film may be prepared and adhered to a second or support layer from which it is removed prior to use, i.e. introduction to the oral cavity. An adhesive may be used to attach the film to the support or backing material which may be any of those known in the art, and is preferably not water soluble. If an adhesive is used, it will desirably be a food grade adhesive that is ingestible and does not alter the properties of the active. Mucoadhesive compositions are particularly useful. The film compositions in many cases serve as mucoadhesives themselves.

The films may be applied under or to the tongue of the mammal. When this is desired, a specific film shape, corresponding to the shape of the tongue may be preferred. Therefore the film may be cut to a shape where the side of the film corresponding to the back of the tongue will be longer than the side corresponding to the front of the tongue. Specifically, the desired shape may be that of a triangle or trapezoid. Desirably, the film will adhere to the oral cavity preventing it from being ejected from the oral cavity and permitting more of the active to be introduced to the oral cavity as the film dissolves.

Another use for the films of the present invention takes advantage of the films' tendency to dissolve quickly when introduce to a liquid. An active may be introduced to a liquid by preparing a film in accordance with the present invention, introducing it to a liquid, and allowing it to dissolve. This may be used either to prepare a liquid dosage form of an active, or to flavor a beverage.

The films of the present invention are desirably packaged in sealed, air and moisture resistant packages to protect the active from exposure oxidation, hydrolysis, volatilization and interaction with the environment. Referring to FIG. 1, a packaged pharmaceutical dosage unit 10, includes each film 12 individually wrapped in a pouch or between foil and/or plastic laminate sheets 14. As depicted in FIG. 2, the pouches 10, 10' can be linked together with tearable or perforated joints 16. The pouches 10, 10' may be packaged in a roll as depicted in FIG. 5 or stacked as shown in FIG. 3 and sold in a dispenser 18 as shown in FIG. 4. The

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dispenser may contain a full supply of the medication typically prescribed for the intended therapy, but due to the thinness of the film and package, is smaller and more convenient than traditional bottles used for tablets, capsules and liquids. Moreover, the films of the present invention 5 dissolve instantly upon contact with saliva or mucosal membrane areas, eliminating the need to wash the dose down with water.

Desirably, a series of such unit doses are packaged together in accordance with the prescribed regimen or 10 treatment, e.g., a 10-90 day supply, depending on the particular therapy. The individual films can be packaged on a backing and peeled off for use.

The features and advantages of the present invention are more fully shown by the following examples which are 15 provided for purposes of illustration, and are not to be construed as limiting the invention in any way.

EXAMPLES

Examples 1-2

Water-soluble thin film compositions were prepared using the amounts described in Table 1. In particular, composition 1 incorporated Vitamin E as an anti-tacking agent along with 25 various other components. Composition 2 contained similar components to composition 1, but absent Vitamin E.

TABLE 1

	Weight (g unless otherwise indicated)	
Component	1	2
Polyethylene oxide	2.8	3.5
Hydroxypropyl cellulose	2.8	3.5
Polydextrose	0.69	0.79
Sucralose	0.35	0.75
Taste-Masking flavor ¹	0.07	0
Titanium dioxide	0.07	0.18
Coated dextromethorphan (45% w/w)	5.56	6.94
Mint flavor	1.26	1.71
Vitamin E ²	3.9	0
WS-3 ³	0.035	0.044
Simethicone emulsion ⁴	0.035	0.09
Water	19.49	32.5
Blue food color	4 drops	5 drops

¹Magna Sweet, available from Mafco Worldwide Corp.

The above components for each composition were combined by mixing until a uniform mixture was achieved, and 28

then cast into films. In particular, the solutions were cast onto release paper (available from Griff Paper & Film) using a K Control Coater with a 350 micron smooth bar. The films were then dried at about 80° C. for about 10 minutes. Composition 1 was dried to a moisture level of about 2.68%. and composition 2 was dried to a moisture level of about

The dried films were tested for various properties, including dissolution testing to determine how long it will take the film to dissolve in the mouth and bend testing to determine flexibility of the film. In addition, a panel observed the tendency of the films to exhibit stickiness in the mouth and the tendency to adhere to the roof of the user's mouth.

To test dissolution rate, an approximately 20 mm by 100 mm piece of film, having a 2.85 g weight attached, was lowered into a 32.5° C. water bath to a depth of about 50 mm. The time required for the film to dissolve and separate into two pieces was determined (in seconds).

The films also were subject to bend testing, i.e., 180° bend test. The dried films were placed in a moisture analyzer (HR73 Moisture Analyzer from Mettler Toledo) to obtain percent moisture and to remove any solvent (e.g. water) remaining in the films after drying at 80° C. in accordance with the present invention. The films then were creased to about 180° and observed for break. Films that broke during creasing were considered a failure. If the film did not break during creasing, a 200 g weight was dropped onto the creased film from a height of about 8.5 mm. Films that broke were considered a failure, and those that did not break were considered a pass. It should be noted, however, that this flexibility test is an extreme test. Films that failed this test are still considered operable within the scope of the present invention. More specifically, there may be certain applications that do not require such extreme flexibility properties.

Both films of compositions 1 and 2 exhibited adequate strength, good tear resistance, passed the 180° bend test both prior and subsequent to placement in the moisture analyzer and dissolved on the tongue at a moderate to fast rate. Composition 1, which contained Vitamin E, exhibited no stickiness in the mouth and did not exhibit a tendency to adhere to the roof of the user's mouth. Composition 2, in contrast, did not contain Vitamin E. Composition 2 exhibited stickiness and tendency to adhere to the roof of the mouth.

Example 3-243

Water-soluble thin films were prepared incorporating silica and magnesium stearate as anti-tacking agents in the amounts described in Table 2. More specifically, various combinations of silica and magnesium stearate were incorporated into a variety of different film compositions as shown in the table below.

TADIE

		IABLE 2		
Example	Film description		Silica ¹ (weight %)	Magnesium stearate (weight %)
3	SOURS		1.5	2.0
4	SOURS		1.5	2.0
5	SOURS		1.5	2.0
6	SOURS		1.5	2.0
7	SOURS		1.5	2.0
8	SOURS		1.5	2.0
9	SOURS		1.5	2.0
10	SOURS		1.5	2.0

²10% solution containing 0.39 g Vitamin E and 3.51 g water

³N-Ethyl-p-menthane-3-carboxamide cooling agent, available from Millennium Chemi-

29TABLE 2-continued

	TABLE 2-continued		
Example	Film description	Silica ¹ (weight %)	Magnesium stearate (weight %)
11	SOURS	1.5	2.0
12	SOURS	1.5	2.0
13	SOURS	1.5	2.0
14	SOURS	1.5	2.0
15	BENZOCAINE/MENTHOL	1.5	1.5
16 17	BENZOCAINE/MENTHOL BENZOCAINE/MENTHOL	1.5 1.5	1.5 1.5
18	BENZOCAINE/MENTHOL BENZOCAINE/MENTHOL	1.5	1.5
19	BENZOCAINE/MENTHOL	1.5	1.5
20	SOURS	2	2.5
21	SOURS	1.5	2
22 23	SOURS SOURS	1.5 1.5	2 2
23	SOURS	1.5	2
25	SOURS	1.5	2
26	SOURS	1.5	2
27	SOURS	1.5	2.5
28 29	SOURS SOURS	1.5 1.5	2.5 2.5
30	SOURS	1.5	2.5
31	SOURS	1.5	2
32	SOURS	1.5	2
33	SOURS	1.5	2.5
34 35	SOURS SOURS	1.5 1.5	2.5 2.5
36	SOURS	1.5	2.5
37	SOURS	1.5	2.5
38	SOURS	1.5	2.5
39 40	SOURS SOURS	1.5 1.5	2.5 2.5
41	ENERGY/WELLNESS SUPPLEMENT ²	1.3	2.3
42	ENERGY/WELLNESS SUPPLEMENT ²	1	2
43	ENERGY/WELLNESS SUPPLEMENT ²	0.9	1
44	ENERGY/WELLNESS SUPPLEMENT ²	1.15	1
45 46	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 0.75	1 1
47	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
48	ENERGY/WELLNESS SUPPLEMENT ²	1	1
49	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1
50 51	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 1	1 1
52	ENERGY/WELLNESS SUPPLEMENT ²	1	1
53	ENERGY/WELLNESS SUPPLEMENT ²	1	1
54	ENERGY/WELLNESS SUPPLEMENT ²	1	1
55 56	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 1	1 1
57	ENERGY/WELLNESS SUPPLEMENT ²	1	1
58	ENERGY/WELLNESS SUPPLEMENT ²	1	1
59	ENERGY/WELLNESS SUPPLEMENT ²	1	1
60 61	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 1	1 1.5
62	ENERGY/WELLNESS SUPPLEMENT ²	0.5	1.5
63	ORAL ANALGESIC	0.54	0.5
64	ORAL ANALGESIC	1.54	1
65 66	ORAL ANALGESIC ORAL ANALGESIC	0.5 1.54	0.54 1
67	ORAL ANALGESIC	1.04	1
68	ORAL ANALGESIC	1.24	1.5
69	ORAL ANALGESIC	1.24	1.5
70 71	ORAL ANALGESIC ORAL ANALGESIC	1.24 1.24	1.5 1.5
72	ORAL ANALGESIC ORAL ANALGESIC	1.24	1.5
73	ORAL ANALGESIC	1.24	1.5
74	ENERGY/WELLNESS SUPPLEMENT ²	0.5	1.5
75 76	MELATONIN MELATONIN	1 1	2 2
77	MELATONIN	1	2
78	MELATONIN	1	1.5
79	MELATONIN	1.1	1.3
80	MELATONIN CUL OPINE DIOVIDE	1.2 1.5	1.3
81 82	CHLORINE DIOXIDE MULTIVITAMIN	1.5	1.5 1
83	MULTIVITAMIN	1	1
84	ZINC/ELDERBERRY	0.5	1
85	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1
86	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1

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TABLE 2-continued

Example	Film description	Silica ¹ (weight %)	Magnesium stearate (weight %)
87	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1
88	MELATONIN	1.1	1.3
89	MULTIVITAMIN	1	1
90 91	B-COMPLEX VITAMIN MULTIVITAMIN	1 1	1 1
92	B-COMPLEX VITAMIN	1	1
93	MULTIVITAMIN	1	1
94	MULTIVITAMIN	1	1
95 96	ENERGY/WELLNESS SUPPLEMENT ² MULTIVITAMIN	0.75 1	1 1
97	MELATONIN	1	1
98	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1
99	ENERGY/WELLNESS SUPPLEMENT ²	0.75	1
100 101	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	0.75 1	1 1.5
101	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
103	ENERGY/WELLNESS SUPPLEMENT ²	1	2
104	ENERGY/WELLNESS SUPPLEMENT ²	1.5	1
105	ENERGY/WELLNESS SUPPLEMENT ²	1	2
106 107	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 1	2 1.5
108	ENERGY/WELLNESS SUPPLEMENT ²	1	2
109	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
110	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
111 112	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1 1	1.5 1.5
113	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
114	ENERGY/WELLNESS SUPPLEMENT ²	1	1.5
115	MULTIVITAMIN	1	1.5
116	MULTIVITAMIN	1	1.5
117 118	IMMUNE BOOSTER MELATONIN	1 1.1	1 1.3
119	MELATONIN	1.1	1.3
120	MELATONIN	1.1	1.3
121	MELATONIN	1.1	1.3
122 123	MELATONIN COLD & COUGH	0.5 1	0.75 1
124	COLD & COUGH	1	1
125	MULTIVITAMIN	1	1
126	MULTIVITAMIN	1	1
127 128	MULTIVITAMIN MULTIVITAMIN	1 1	1.5 1
129	MULTIVITAMIN	1	1
130	MULTIVITAMIN	1	1.5
131	MULTIVITAMIN	1	1
132 133	MULTIVITAMIN B-COMPLEX VITAMIN	1 1	1 1
134	B-COMPLEX VITAMIN	1	1
135	B-COMPLEX VITAMIN	1	1
136	B-COMPLEX VITAMIN	1	1
137 138	ENERGY/WELLNESS SUPPLEMENT ² ENERGY/WELLNESS SUPPLEMENT ²	1.5 1.5	1.5 1.5
139	ENERGY/WELLNESS SUPPLEMENT ²	1.5	1.5
140	MULTIVITAMIN	1	1
141	B-COMPLEX VITAMIN	1	1
142	B-COMPLEX VITAMIN	1 1	1 1
143 144	MULTIVITAMIN B-COMPLEX VITAMIN	1	1
145	MULTIVITAMIN	1	1
146	MULTIVITAMIN	1	1
147	MULTIVITAMIN	1	1
148 149	MULTIVITAMIN B-COMPLEX VITAMIN	1 1	1 1
150	B-COMPLEX VITAMIN	1	1
151	MULTIVITAMIN	1	1
152	MULTIVITAMIN	1	1
153 154	MULTIVITAMIN MULTIVITAMIN	1 1.5	1 0.3
155	ENERGY/WELLNESS SUPPLEMENT ²	1.5	1
156	ENERGY/WELLNESS SUPPLEMENT ²	1	1
157	ENERGY/WELLNESS SUPPLEMENT ²	1	1
158 159	MULTIVITAMIN MULTIVITAMIN	1 1	1 1
160	MULTIVITAMIN MULTIVITAMIN	1.5	0.3
161	MULTIVITAMIN	1	1
162	B-COMPLEX VITAMIN	1	1

33TABLE 2-continued

	TABLE 2-continued		
Example	Film description	Silica ¹ (weight %)	Magnesium stearate (weight %)
163	B-COMPLEX VITAMIN	1	1
164	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
165	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
166	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
167	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
168	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
169	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
170	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
171	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
172 173	ENERGY/WELLNESS SUPPLEMENT ² MULTIVITAMIN	1 1	0.5 1
173	MULTIVITAMIN	1	1
175	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
176	MULTIVITAMIN	1	1
177	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
178	ENERGY/WELLNESS SUPPLEMENT ²	0.5	0.5
179	MULTIVITAMIN	1	1
180	MULTIVITAMIN	1	1
181	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
182	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
183	MULTIVITAMIN	1	1
184 185	MULTIVITAMIN MULTIVITAMIN	1 1	1 1
186	MULTIVITAMIN	1	1
187	MULTIVITAMIN	1	1
188	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
189	MULTIVITAMIN	1	1
190	MULTIVITAMIN	1	1
191	MULTIVITAMIN	1	1
192	MULTIVITAMIN	1	1
193	MULTIVITAMIN	1.37	2.05
194	MULTIVITAMIN	1	1
195	MULTIVITAMIN	1	1
196	MULTIVITAMIN	1	1
197 198	B-COMPLEX VITAMIN ENERGY/WELLNESS SUPPLEMENT ²	1 1	1 0.5
198	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
200	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
201	B-COMPLEX VITAMIN	1	1
202	MULTIVITAMIN	1	1
203	MULTIVITAMIN	1	1
204	MELATONIN	1.1	1.3
205	MULTIVITAMIN	1.5	0.3
206	MULTIVITAMIN	1	1
207	STRESS RELIEF	1	0.3
208 209	MULTIVITAMIN MULTIVITAMIN	1 1	1 1
210	MULTIVITAMIN	1	1
211	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
212	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
213	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
214	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
215	ENERGY/WELLNESS SUPPLEMENT ²	1	0.5
216	MULTIVITAMIN	1.5	0.3
217	MELATONIN	1	0.5
218	MELATONIN GENERAL REF	1	0.5
219 220	STRESS RELIEF MULTIVITAMIN	1 1	0.3 1
221	MELATONIN	1	0.5
222	MULTIVITAMIN	1.5	0.3
223	MULTIVITAMIN	1	1
224	MULTIVITAMIN	1	1
225	CINNAMINT	1	1
226	MELATONIN	1	0.5
227	MELATONIN	1	0.5
228	B-COMPLEX VITAMIN	1	1
229	MULTIVITAMIN	1	1
230 231	MULTIVITAMIN MULTIVITAMIN	1 1	1 1
231	MULTIVITAMIN MULTIVITAMIN	1	1
232	MULTIVITAMIN	1	1
234	MULTIVITAMIN	1	1
235	MULTIVITAMIN	1	1
236	MULTIVITAMIN	1	1
237	MULTIVITAMIN	1	1
238	MULTIVITAMIN	1	1

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Example	Film description	Silica ¹ (weight %)	Magnesium stearate (weight %)
239	MULTIVITAMIN	1	1
240	MULTIVITAMIN	1	1
241	BENZOCAINE/MENTHOL	1.5	1.5
242	MULTIVITAMIN	1	1
243	DEXTROMETHORPHAN HYDROBROMIDE	0.5	1.82

¹Sipernat 500LS, available from Degussa

In addition to silica and magnesium stearate, each of the films listed above contains a variety of components, such as polymers and flavors, among others. The remainder of the components are provided below for each film description used in Table 2.

Films identified in Table 2 above as "SOURS" contain the $_2$ following components listed in Table 3:

TABLE 3

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-60%
CITRIC ACID	0.01%-40%
NATURAL &ARTIFICIAL FLAVORS	0.01%-25%
GUM ARABIC	0.01%-10%
MAGNESIUM STEARATE	0.01%-10%
SODIUM HEXAMETAPHOSPHATE	0.01%-5%
SILICA	0.01%-2%
POLYSORBATE 80	0%-5%
MALIC ACID	0.01%-10%
ASPARTAME	0.01%-3.5%
POTASSIUM ACESULFAME	0.01%-0.5%
DYE	0.01%-1%
POTASSIUM SORBATE	0.01%-0.1%
SODIUM BENZOATE	0.01%-0.1%

Films identified in Table 2 above as "Benzocaine/Menthol" contain the following components listed in Table 4:

TABLE 4

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
NATURAL &ARTIFICIAL FLAVORS	0.01%-25%
POLYETHYLENE OXIDE	0.01%-50%
MENTHOL CRYSTALS	0.01%-30%
CORN STARCH	0.01%-30%
BENZOCAINE	0.01%-10%
SUCRALOSE	0.01%-5%
MALIC ACID	0.01%-5%
MAGNESIUM STEARATE	0.01%-10%
SILICA	0.01%-2%
TITANIUM DIOXIDE	0.01%-5%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
DYE	0.01%-1%

Films identified in Table 2 above as "Energy/Wellness Supplement" contain the following components listed in Table 5:

TABLE 5

Component	Weight %	
HYDROXYPROPYL METHYLCELLULOSE	0%-70%%	
HYDROXYPROPYL CELLULOSE	0%-40%	
PECTIN	0%-40%	

TABLE 5-continued

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Component	Weight %
NATURAL &ARTIFICIAL FLAVORS/FLAVOR ADJUVANTS	0%-30%
POLYDEXTROSE	0.01%-30%
SODIUM CARBOXYMETHYLCELLULOSE	0%-10%
ENERGY/WELLNESS ACTIVES ²	0.01%-50%
ERYTHRITOL	0%-20%
SUCRALOSE	0.01%-5%
CITRIC ACID	0%-10%
MAGNESIUM STEARATE	0.01%-10%
GLYCERYL MONOOLEATE	0%-1%
SILICA	0.01%-2%
POLYSORBATE 80	0%-1%
SORBITAN MONOOLEATE	0%-1%
POTASSIUM SORBATE	0%-0.1%
SODIUM BENZOATE	0%-0.1%
SODIUM HEXAMETAPHOSPHATE	0%-10%
PROPYLENE GLYCOL	0%-25%
GUM ARABIC	0%-10%
DYE	0.01%-1%

Films identified in Table 2 above as "ORAL ANALGE-SIC" contain the following components listed in Table 6:

TABLE 6

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
CHOLINE SALICYLATE	0.01%-60%
NATURAL &ARTIFICIAL FLAVORS	0.01%-10%
MAGNESIUM STEARATE	0.01-5%
SILICA	0.01-2%
CETALKONIUM CHLORIDE	0.01%-5%
METHYL PARABEN	0.01%-0.1%
DIMETHYLPOLYSILOXANE	0.01%-0.05%

Films identified in Table 2 above as "Melatonin" contain the following components listed in Table 7:

TABLE 7

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
NATURAL &ARTIFICIAL FLAVORS/	0.01%-20%
FLAVOR ADJUVANTS	
POLYETHYLENE OXIDE	0.01%-30%
MELATONIN	0.01%-20%
PECTIN	0.01%-10%
POLYDEXTROSE	0.01%-20%
SUCRALOSE	0.01%-5%
MAGNESIUM STEARATE	0.01%-10%
SILICA	0.01%-2%
GLYCERYL MONOOLEATE	0.01%-1%
FITANIUM DIOXIDE	0.01%-5%
MONOAMMONIUM GLYCYRRHIZINATE	0.01%-2%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
DYE	0.01%-1%

²Energy/Wellness Supplement may contain any/all of the following actives or combinations thereof: Green Tea, Guarana, Chromium Picolinate, Caffeine, Yohimbie HCl, Taurine, Vitamin B3, Vitamin B6, Vitamin B12

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Films identified in Table 2 above as "Chlorine Dioxide" contain the following components listed in Table 8:

TABLE 8

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
POLYETHYLENE OXIDE	0.01%-50%
POLYDEXTROSE	0.01%-20%
NATURAL &ARTIFICIAL FLAVORS/	0.01%-30%
FLAVOR ADJUVANTS	
MAGNESIUM STEARATE	0.01%-5%
SILICA	0.01%-2%
SUCRALOSE	0.01%-5%
ZINC GLUCONATE DIHYDRATE	0.01%-5%
CITRIC ACID	0.01%-2%
GLYCERYL MONOOLEATE	0.01%-1%
SODIUM HYDROXIDE	0.01%-5%
SODIUM BICARBONATE	0.01%-5%
CHLORINE DIOXIDE 2% SOLUTION	0.01%-10%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
DYE	0.01%-1%
SODIUM BENZOATE	0.01%-0.1%

Films identified in Table 2 above as "Multivitamin" contain the following components listed in Table 9:

TABLE 9

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-50%
NATURAL &ARTIFICIAL FLAVORS	0.01%-20%
NIACINAMIDE - 100% (Vitamin B3)	0.01%-30%
POLYETHYLENE OXIDE	0.01%-30%
POLYDEXTROSE	0.01%-20%
ASCORBIC ACID - 100% (Vitamin C)	0.01%-20%
50% VITAMIN E ACETATE - 91.2%	0.01%-10%
CALCIUM d-PANTOTHENATE - 92% (Vitamin B5)	0.01%-10%
SUCRALOSE	0.01%-5%
VITAMIN A PALMITATE - 15%	0.01%-10%
PYRIDOXINE HYDROCHLORIDE - 82.3%	0.01%-10%
(VITAMIN B6)	
RIBOFLAVIN - 100% (Vitamin B2)	0.01%-10%
THIAMINE HYDROCHLORIDE - 89.2%	0.01%-10%
(Vitamin B1)	
MAGNESIUM STEARATE	0.01%-2%
SILICA	0.01%-2%
GLYCERYL MONOOLEATE	0.01%-1%
5% VITAMIN K - 100%	0.01%-5%
2.5% VITAMIN D3 LIQUID - 100%	0.01%-5%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
CYANOCOBALAMIN - 100% (Vitamin B12)	0.001%-1%

Films identified in Table 2 above as "Zinc/Elderberry" contain the following components listed in Table 10:

TABLE 10

Component	Weight %	
HYDROXYPROPYL METHYLCELLULOSE	0.01%-60%	
ZINC GLUCONATE	0.01%-20%	55
ELDERBERRY EXTRACT	0.01%-20%	
FRUCTOSE	0.01%-20%	
NATURAL & ARTIFICIAL FLAVORS	0.01%-30%	
POLYETHYLENE OXIDE	0.01%-20%	
POLYDEXTROSE	0.01%-20%	
ASCORBIC ACID - 100% (Vitamin C)	0.01%-20%	60
SUCRALOSE	0.01%-5%	
GLYCERYL MONOOLEATE	0.01%-1%	
MAGNESIUM STEARATE	0.01%-5%	
TITANIUM DIOXIDE	0.01%-2%	
SILICA	0.01%-2%	
BUTYLATED HYDROXYTOLUENE	0.01%-1%	65

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Films identified in Table 2 above as "B-Complex Vitamin" contain the following component listed in Table 11:

TABLE 11

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-60%
POLYETHYLENE OXIDE	0.01%-50%
CALCIUM d-PANTOTHENATE - 92% (Vitamin	0.01%-20%
B5)	
POLYDEXTROSE	0.01%-30%
NATURAL & ARTIFICIAL FLAVORS/FLAVOR	0.01%-25%
ADJUVANTS	
PYRIDOXINE HYDROCHLORIDE - 82.3%	0.01%-20%
(VITAMIN B6)	
RIBOFLAVIN - 100% (Vitamin B2)	0.01%-20%
5 THIAMINE HYDROCHLORIDE - 89.2%	0.01%-20%
(Vitamin B1)	
SUCRALOSE	0.01%-5%
PROPYLENE GLYCOL	0.01%-5%
MAGNESIUM STEARATE	0.01%-10%
SILICA	0.01%-2%
GLYCERYL MONOOLEATE	0.01%-1%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
DYE	0.01%-1%
CYANOCOBALAMIN - 100% (Vitamin B12)	0.001%-1%

Films identified in Table 2 above as "Immune Boostee" contain the following components listed in Table 12:

TABLE 12

Component	Weight %
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
POLYDEXTROSE	0.01%-50%
POLYETHYLENE OXIDE	0.01%-50%
ZINC CITRATE TRIHYDRATE	0.01%-40%
SUCRALOSE	0.01%-5%
NATURAL FLAVORS	0.01%-20%
CITRIC ACID	0.01%-20%
MAGNESIUM STEARATE	0.01%-10%
SILICA	0.01%-2%
SODIUM CITRATE	0.01%-5%
GLYCERYL MONOOLEATE	0.01%-1%
BUTYLATED HYDROXYTOLUENE	0.01%-2%
MONOAMMONIUM GLYCYRRHIZINATE	0.01%-1%
DYE	0.01%-1%

Films identified in Table 2 above as "Cold & Cough" contain the following components listed in Table 13:

TABLE 13

	Component	Weight %
	HYDROXYPROPYL METHYLCELLULOSE	0.01%-60%
	POLYDEXTROSE	0.01%-30%
	NATURAL &ARTIFICIAL FLAVORS	0.01%-25%
	POLYETHYLENE OXIDE	0.01%-50%
	ASCORBIC ACID - 100% (Vitamin C)	0.01%-30%
	ZINC CITRATE DIHYDRATE	0.01%-20%
ECHINACEA PURPURE SUCRALOSE	ECHINACEA PURPUREA	0.01%-20%
	SUCRALOSE	0.01%-10%
	PECTIN	0.01%-20%
	CITRIC ACID	0.01%-10%
	SODIUM CITRATE	0.01%-5%
	MAGNESIUM STEARATE	0.01%-10%
	SILICA	0.01%-2%
	GLYCERYL MONOOLEATE	0.01%-1%
	DYE	0.01%-1%
	BUTYLATED HYDROXYTOLUENE	0.01%-1%
	MONOAMMONIUM GLYCYRRHIZINATE	0.01%-1%

Films identified in Table 2 above as "Stress Relief" contain the following components listed in Table 14:

39 TABLE 14

40 TABLE 17

Component	Weight %	
HYDROXYPROPYL METHYLCELLULOSE	0.01%-60%	
CHAMOMILE	0.01%-40%	
PASSION FLOWER	0.01%-40%	
PECTIN	0.01%-20%	
NATURAL &ARTIFICIAL FLAVORS	0.01%-25%	
GLYCERIN	0.01%-10%	
POLYSORBATE 80	0%-2%	
SUCRALOSE	0.01%-5%	
POLYDIMETHYLSILOXANE EMULSION	0.01%-2%	
ASPARTAME	0.01%-5%	
POTASIUM ACESULFAME	0.01%-3%	
POTASSIUM SORBATE	0.01%-1%	

Films identified in Table 2 above as "Cinnamint" contain the following components listed in Table 15:

TABLE 15

Component	Weight %	
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%	
POLYETHYLENE OXIDE	0.01%-50%	
POLYDEXTROSE	0.01%-30%	
BUTYLATED HYDROXYTOLUENE	0.01%-1%	
GLYCERYL MONOOLEATE	0.01%-1%	
MAGNESIUM STEARATE	0.01%-10%	
SILICA	0.01%-2%	
POTASSIUM SORBATE	0.01%-0.1%	
SODIUM BENZOATE	0.01%-0.1%	
NATURAL &ARTIFICIAL FLAVORS	0.01%-30%	
SUCRALOSE	0.01%-5%	
XYLITOL	0.01%-10%	
DYE	0.01%-1%	

Films identified in Table 2 above as "Dextromethorphan 35 Hydrobromide" contain the following components listed in Table 16:

TABLE 16

Component	Weight %
Dextromethorphan Hydrobromide 60%	0.01%-60%
POLYETHYLENE OXIDE	0.01%-70%
POLYDEXTROSE	0.01%-30%
HYDROXYPROPYL METHYLCELLULOSE	0.01%-70%
NATURAL &ARTIFICIAL FLAVORS	0.01%-30%
SUCRALOSE	0.01%-5%
MAGNESIUM STEARATE	0.01%-10%
SILICA	0%-2%
SODIUM BICARBONATE	0.01%-5%
XANTHAN GUM	0.01%-10%
TITANIUM DIOXIDE	0.01%-5%
BUTYLATED HYDROXYTOLUENE	0.01%-1%
DYE	0.01%-1%

The films prepared in these Examples exhibited improved glidant properties, particularly the ability to slide against one another without sticking together.

Examples 244-300

Water-soluble thin films were prepared incorporating silica and magnesium stearate as anti-tacking agents in the amounts described in Table 17. More specifically, various combinations of silica and magnesium stearate were incorporated into a variety of different film compositions as shown in the table below.

		1	Magnesium
Evennle	Film description	Silica ¹	stearate (weight
Example	Film description	(weight %)	%)
244	SOURS	1.5	2.5
245	ENERGY/WELLNESS	0.75	1
	SUPPLEMENT ²		
246	ENERGY/WELLNESS	0.75	1
2.17	SUPPLEMENT ²		
247	MELATONIN	1	2 1.5
248 249	MELATONIN CHLORINE DIOXIDE	1.5 1.5	1.5
250	MELATONIN	1.5	1.5
251	MELATONIN	1.5	1.5
252	MELATONIN	1.5	1.5
253	MELATONIN	1.5	1.5
254	CHLORINE DIOXIDE	1.5	1.5
255	MELATONIN	1.1	1.3
256	MULTIVITAMIN	1	1
257	MULTIVITAMIN	1	1
258	B COMPLEX VITAMIN	1	1
259	MULTIVITAMIN	1	1
260 261	B COMPLEX VITAMIN	1 1	1 1
262	COLD & COUGH MULTIVITAMIN	1	1
263	MULTIVITAMIN	1	1
264	MULTIVITAMIN	1	1
265	MULTIVITAMIN	*	•
266	MULTIVITAMIN		
267	MULTIVITAMIN		
268	MULTIVITAMIN	1	1.5
269	IMMUNE BOOSTER	1.16	1.16
270	MULTIVITAMIN	1	1
271	ENERGY/WELLNESS	1	1.5
272	SUPPLEMENT ²	1	1.5
272 273	MULTIVITAMIN MULTIVITAMIN	1 1	1.5 1.5
273	MELATONIN	1.1	1.3
275	MULTIVITAMIN	1	1
276	MULTIVITAMIN	1	1
277	MULTIVITAMIN	1	1
278	MULTIVITAMIN	1	1
279	MULTIVITAMIN	1	1
280	ENERGY/WELLNESS	1.5	1.5
	SUPPLEMENT ²		
281	MULTIVITAMIN	1	1.5
282	MULTIVITAMIN	1	1.5
283	MULTIVITAMIN	1.5	0.3
284 285	MULTIVITAMIN B-COMPLEX VITAMIN	1 1	1 1
286	ENERGY/WELLNESS	1	0.5
200	SUPPLEMENT ²	1	0.5
287	MULTIVITAMIN	1	1
288	MELATONIN	1.1	1.3
289	B-COMPLEX VITAMIN	1	1
290	B-COMPLEX VITAMIN	1	1
291	ENERGY/WELLNESS	1	0.5
	SUPPLEMENT ²		
292	ENERGY/WELLNESS	1	0.5
200	SUPPLEMENT ²		
293	B-COMPLEX VITAMIN	1	1
294	MULTIVITAMIN	1	1
295 296	CHLORINE DIOXIDE MULTIVITAMIN	1.5 1	1.5 1.5
296 297	MULTIVITAMIN	1	1.5
298	MULTIVITAMIN	1	1
299	MULTIVITAMIN	1	1
300	BENZOCAINE/MENTHOL	1.5	1.5

¹Sipernat 500LS, available from Degussa

Besides silica and magnesium stearate, the remainder of the components contained in the films listed in Table 17 are provided in connection with Table 2 above. The film descriptions used in Tables 2 and 17 are the same.

The films prepared in these Examples exhibited improved glidant properties, particularly the ability to slide against one another without sticking together.

²Energy/Wellness Supplement may contain any/all of the following actives or combinations thereof: Green Tea, Guarana, Chromium Picolinate, Caffeine, Yohimbie HCl, Taurine, Vitamin B3, Vitamin B6, Vitamin B12

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While there have been described what are presently believed to be the preferred embodiments of the invention, those skilled in the art will realize that changes and modifications may be made thereto without departing from the spirit of the invention, and it is intended to include all such 5 changes and modifications as fall within the true scope of the invention.

What is claimed is:

- 1. A continuously cast film for delivery of an active in individual self-supporting oral unit doses, said individual 10 self-supporting oral unit doses cut from said continuously cast film containing a desired amount of said active, said continuously cast film comprising:
 - an ingestible, water-soluble, polymer matrix;
 - at least one anti-tacking agent selected from the group 15 consisting of stearate; stearic acid; vegetable oil; wax; a blend of magnesium stearate and sodium lauryl sulfate; boric acid; surfactant; sodium benzoate; sodium acetate; sodium chloride; DL-Leucine; polyethylene glycol; sodium oleate; sodium lauryl sulfate; 20 magnesium lauryl sulfate; talc; corn starch;
 - amorphous silicon dioxide; silicon dioxide; metallic stearate; Vitamin E; Vitamin E TPGS; silica and combinations thereof; and
 - a substantially uniform distribution of said active sub- 25 stantially locked-in within said polymer matrix, wherein said active is selected from the group consisting of cosmetic agents,
 - pharmaceutical agents, vitamins, bioactive agents and combinations thereof wherein said active is substan- 30 tially uniformly distributed in said continuously cast film, whereby said substantially uniform distribution is measured by substantially equally sized individual self supporting oral unit doses cut from said continuously desired amount of said active.
- 2. The continuously cast film of claim 1, wherein said anti-tacking agent is present in an amount of about 0.01% to about 20% by weight of said film.
- 3. The continuously cast film of claim 1, wherein said 40 water-soluble polymer matrix comprises a polymer selected from the group consisting of cellulose, cellulose derivatives, polyethylene oxide (PEO), pullulan, hydroxypropylmethyl cellulose (HPMC), hydroxyethyl cellulose (HEC), hydroxypropyl cellulose, polyvinyl pyrrolidone (PVP), carboxym- 45 ethyl cellulose, polyvinyl alcohol, polysaccharides, sodium alginate, xanthan gum, tragancanth gum, guar gum, acacia gum, arabic gum, polyacrylic acid, methylmethacrylate copolymer, carboxyvinyl copolymers, starch, and combinations thereof.
- 4. The continuously cast film of claim 1, wherein said active is selected from the group consisting of anxiolytics, central nervous system stimulants, ace-inhibitors, antianginal drugs, anti-arrhythmias, anti-asthmatics, anti-cholesterolemics, anesthetics, anti-convulsants, anti-depressants, 55 anti-diabetic agents, anti-diarrhea preparations, antidotes, anti-histamines, anti-hypertensive drugs, anti-inflammatory agents, anti-lipid agents, anti-manics, anti-nauseants, antistroke agents, anti-thyroid preparations, anti-tumor drugs, anti-viral agents, acne drugs, alkaloids, amino acid prepa- 60 rations, anti-tussives, anti-uricemic drugs, anti-viral drugs, anabolic preparations, systemic and non-systemic anti-infective agents, anti-neoplastics, anti-parkinsonian agents, anti-rheumatic agents, appetite stimulants, biological response modifiers, blood modifiers, bone metabolism regulators, cardiovascular agents, cholinesterase inhibitors, contraceptives, decongestants, dietary supplements, dopamine

- receptor agonists, endometriosis management agents, enzymes, erectile dysfunction therapies, fertility agents, gastrointestinal agents, homeopathic remedies, hormones, hypercalcemia and hypocalcemia management agents, immunomodulators, immunosuppressives, migraine preparations, motion sickness treatments, muscle relaxants, obesity management agents, osteoporosis preparations, oxytoparasympatholytics, parasympathomimetics, prostaglandins, psychotherapeutic agents, respiratory agents, sedatives, smoking cessation aids, sympatholytics, tremor preparations, urinary tract agents, vasodilators, laxatives, antacids, ion exchange resins, anti-pyretics, appetite suppressants, expectorants, anti-anxiety agents, anti-ulcer agents, anti-inflammatory substances, coronary dilators, cerebral dilators, peripheral vasodilators, psycho-tropics, stimulants, anti-hypertensive drugs, vasoconstrictors, migraine treatments, antibiotics, tranquilizers, anti-psychotics, anti-tumor drugs, anti-coagulants, anti-thrombotic drugs, hypnotics, anti-emetics, anti-nauseants, anti-convulsants, neuromuscular drugs, hyper- and hypo-glycemic agents, thyroid and anti-thyroid preparations, diuretics, antispasmodics, uterine relaxants, anti-obesity drugs, erythropoietic drugs, anti-asthmatics, cough suppressants, mucolytics, DNA and genetic modifying drugs, and combinations thereof.
- 5. The continuously cast film of claim 1, wherein said active is selected from the group consisting of an opiate, an opiate derivative, an analgesic, a biological response modifier, a urinary tract agent, tadalafil, apomorphine, a migraine treatment, a hormone, an anti-convulsant, alprazolam and combinations thereof.
- 6. A continuously cast film for mucosal delivery of an active in individual self-supporting oral unit doses, said individual self-supporting oral unit doses cut from said cast film which do not vary by more than 10% of said 35 continuously cast film containing a desired amount of said active, said continuously cast film comprising;
 - (a) at least one first film layer comprising:
 - an ingestible, water-soluble polymer matrix; and
 - (b) at least one additional film layer comprising:
 - an ingestible, water-soluble polymer matrix; and
 - wherein the water-soluble polymer matrix of one or more of the first and additional film layers includes a substantially uniform distribution of said active substantially locked-in within said polymer matrix, wherein said active is selected from the group consisting of cosmetic agents, pharmaceutical agents, vitamins, bioactive agents and combinations thereof; and
 - wherein one or more of the first and additional film layers include at least one anti-tacking agent selected from the group consisting of stearate; stearic acid; vegetable oil; wax; a blend of magnesium stearate and sodium lauryl sulfate; boric acid; surfactant; sodium benzoate;
 - sodium acetate; sodium chloride; DL-Leucine; polyethylene glycol; sodium oleate; sodium lauryl sulfate; magnesium lauryl sulfate; talc; corn starch; amorphous silicon dioxide; silicon dioxide; metallic stearate; Vitamin E; Vitamin E TPGS; silica and combinations thereof; and
 - wherein said active is substantially uniformly distributed in said continuously cast film, whereby said substantially uniform distribution is measured by substantially equal sized individual self-supporting oral unit doses cut from said continuously east film which do not vary by more than 10% of said desired amount of said active.
 - 7. The continuously cast film of claim 6, wherein said active is selected from the group consisting of an opiate, an

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opiate derivative, an analgesic, a biological response modifier, a urinary tract agent, tadalafil, an anti-convulsant, apomorphine, a migraine treatment, a hormone, alprazolam and combinations thereof.

- **8**. A continuously cast film for delivery of an active in 5 individual self-supporting oral unit doses, said individual self-supporting oral unit doses cut from said continuously cast film containing a desired amount of said active, said continuously cast film comprising:
 - an ingestible, water-soluble polymer matrix comprising a polymer selected from the group consisting of cellulose, a cellulose derivative, polyethylene oxide (PEO), pullulan, hydroxpopylmethyl cellulose (HPMC), hydroxyethyl cellulose (HEC), hydroxypropyl cellulose, polyvinyl pyrrolidone (PVP), carboxymethyl cellulose, polyvinyl alcohol, polys accharide, sodium alginate, xanthan gum, tragancanth gum, guar gum, acacia gum, arabic gum, polyacrylc acid, methylmethacrylate copolymer, carboxyvinyl copolymers, starch, gelatin, and combinations thereof:
 - at least one anti-tacking agent selected from the group consisting of stearate; stearic acid; vegetable oil; wax; a blend of magnesium stearate and sodium lauryl sulfate; boric acid:
 - surfactant; sodium benzoate; sodium acetate; sodium 25 chloride; DL-Leucine; polyethylene glycol; sodium oleate; sodium lauryl sulfate; magnesium lauryl sulfate; talc; corn starch;
 - amorphous silicon dioxide; silicon dioxide; metallic stearate; Vitamin E; Vitamin E TPGS; silica and combinations thereof;
 - a substantially uniform distribution of said active substantially locked-in within said polymer matrix, wherein said active is selected from the group consisting of pharmaceutical agents, bioactive agents and 35 combinations thereof;
 - wherein said active is substantially uniformly distributed in said continuously cast film, whereby said substantially uniform distribution is measured by substantially equally sized individual self-supporting oral unit doses 40 cut from said continuously cast film which do not vary by more than 10% of said desired amount of said active.
- **9**. The continuously cast film of claim **8**, further comprising a component selected from the group consisting of citric 45 acid, propylene glycol, a sweetener, a preservative, a coloring agent, a flavor, a flavor enhancer and combinations thereof.
- 10. The continuously cast film of claim 8, wherein the film is a multi-layered film.
- 11. The continuously cast film of claim 8, wherein said active is selected from the group consisting of an opiate, an opiate derivative, an analgesic, a biological response modifier, a urinary tract agent, tadalafil, an anti-convulsant, apomorphine, a migraine treatment, a hormone, alprazolam 55 and combinations thereof.
- 12. The continuously cast film of claim 9, wherein the flavor is selected from the group of mint oil, citrus flavor and combinations thereof.
- 13. The continuously cast film of claim 8, further comprising a buffer.
- 14. The continuously cast film of claim 9, wherein the sweetener is selected form the group consisting of acesulfame K, sodium saccharin, aspartame and combinations thereof.
- 15. The continuously cast film of claim 8, further comprising an inorganic pigment or filler.

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- 16. The continuously cast film of claim 1, wherein the individual unit doses contain a solvent selected from the group consisting of water, ethanol, isopropanol, acetone, methylene chloride and combinations thereof, present in an amount not greater than 10% by weight.
- 17. The continuously cast film of claim 1, further comprising polyethylene oxide and an anti-tacking agent.
- 18. The continuously cast film of claim 1, wherein the individual self-supporting oral unit doses further comprise a layer which is selected from the group consisting of a cast layer, a laminated layer, a coated layer and combinations thereof.
- 19. The continuously cast film of claim 18, wherein the layer contains an active.
- 20. The continuously cast film of claim 18, wherein the layer does not contain an active.
- 21. The continuously cast film of claim 6, wherein the individual unit doses contain a solvent selected from the group consisting of water, ethanol, isopropanol, acetone, methylene chloride and combinations thereof, present in an amount not greater than 10% by weight.
 - 22. The continuously cast film of claim 6, wherein the sodium benzoate content is from about 0.01 to about 20% by weight of the cast film.
 - 23. The continuously cast film of claim 8, wherein the individual unit doses contain a solvent selected from the group consisting of water, ethanol, isopropanol, acetone, methylene chloride and combinations thereof, present in an amount not greater than 10% by weight.
 - 24. The continuously cast film of claim 8, further comprising polyethylene oxide and an anti-tacking agent.
 - 25. The continuously cast film of claim 24, wherein the anti-tacking agent is selected from the group consisting of silicon dioxide, amorphous silicon dioxide, and combinations thereof.
 - 26. A continuously cast film for delivery of an active in individual self-supporting oral unit doses, said individual self-supporting oral unit doses cut from said continuously cast film containing a desired amount of said active, said continuously cast film comprising:
 - an ingestible, water-soluble, polymer matrix comprising a polymer selected from the group consisting of hydroxyethylcellulose, hydroxypropylcellulose, carboxymethyl cellulose, polyacrylic acid and combinations thereof;
 - said active being selected from the group consisting of an opiate, an opiate derivative and combinations thereof; sodium benzoate;

citric acid;

propylene glycol;

vitamin E acetate;

an inorganic pigment or filler;

- a flavor selected from the group consisting of citrus flavor,
- peppermint oil and combinations thereof; a sweetener selected from the group consisting of acesulfame K, aspartame, sodium saccharin and combinations thereof:
- wherein said active is substantially uniformly distributed in said continuously cast film, whereby said substantially uniform distribution is measured by substantially equally sized individual self-supporting oral unit doses cut from said continuously cast film which do not vary by more than 10% of said desired amount of said active.
- 27. The continuously cast film of claim 26, wherein the individual unit doses contain a solvent selected from the group consisting of water, ethanol, isopropanol, acetone,

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methylene chloride and combinations thereof, present in an amount not greater than 10% by weight.

- **28**. The continuously cast film of claim **26**, wherein the individual self-supporting oral unit doses further comprise a layer which is selected from the group consisting of a cast 5 layer, a laminated layer, a coated layer and combinations thereof.
- 29. The continuously cast film of claim 28, wherein the layer contains an active.
- **30**. The continuously cast film of claim **28**, wherein the 10 layer does not contain an active.

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